

$$X_j(t) = \sum_n X_j(n,t) = \frac{1}{m_j^{1/2}} \sum_n p_j(n) X(n,t)$$



Technologies for Deep Decarbonization Using High Temperature Liquid Metal

Asegun Henry, Ph.D.

Noyce Career Development Chair

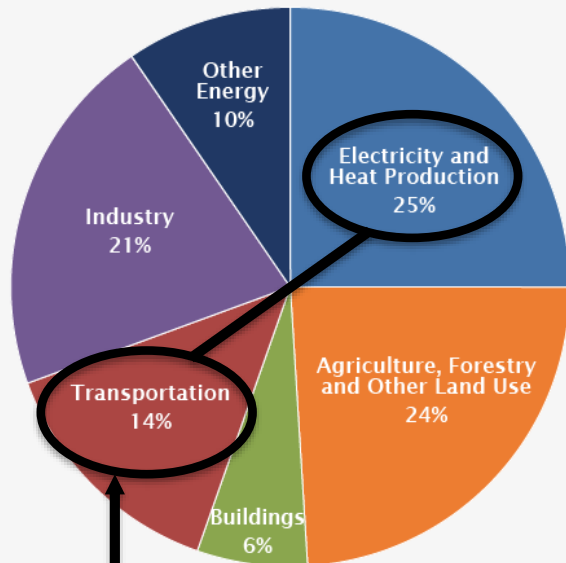
Associate Professor

Department of Mechanical Engineering
Massachusetts Institute of Technology



Energy Storage Impact

Global Greenhouse Gas Emissions
by Economic Sector



Energy storage is the key to decarbonizing electricity and transportation

Five thermal energy grand challenges for decarbonization

Roughly 90% of the world's energy use today involves generation or manipulation of heat over a wide range of temperatures. Here, we note five key applications of research in thermal energy that could help make significant progress towards mitigating climate change at the necessary scale and urgency.

Asegun Henry, Ravi Prasher and Arun Majumdar

Thermal storage systems

As solar and wind electricity penetration has increased, its intermittency has hastened the need for low-cost storage over a wide range of time scales, from seconds to days, and even seasonal storage. Current technologies, such as pumped hydroelectricity, are geographically limited and lithium-ion batteries (~US\$80–100 kWh⁻¹ capital cost) are too expensive for the multi-day storage targets (~US\$3–30 kWh⁻¹) needed to fully decarbonize the grid^{1,2}. Solving this problem could enable full decarbonization of the grid, thereby reducing global GHG emissions by ~25%^{3,4}. Thus, the storage problem is one of the single most impactful problems to be solved.

Several new thermal energy storage (TES) concepts have been proposed^{5,6}. While it is relatively easy to convert electricity to

ev (or batteries), nonetheless, since the specific heat of virtually all materials is the same on a molar basis, at high temperatures, TES can make use of extremely abundant and low-cost materials that are impure or even recycled.

Although several embodiments of TES have been put forth, they are still early stage and have not yet reached commercial deployment. Thus, there is a need to continue developing more competing embodiments that exploit other thermal storage materials and mechanisms. In particular, it is of utmost importance to develop full-system concepts that carefully consider all of the practical issues (for example, materials degradation and compatibility over time, safety, system integration, transients and so on) that might stifle or prevent commercial deployment. For example, systems that utilize a liquid

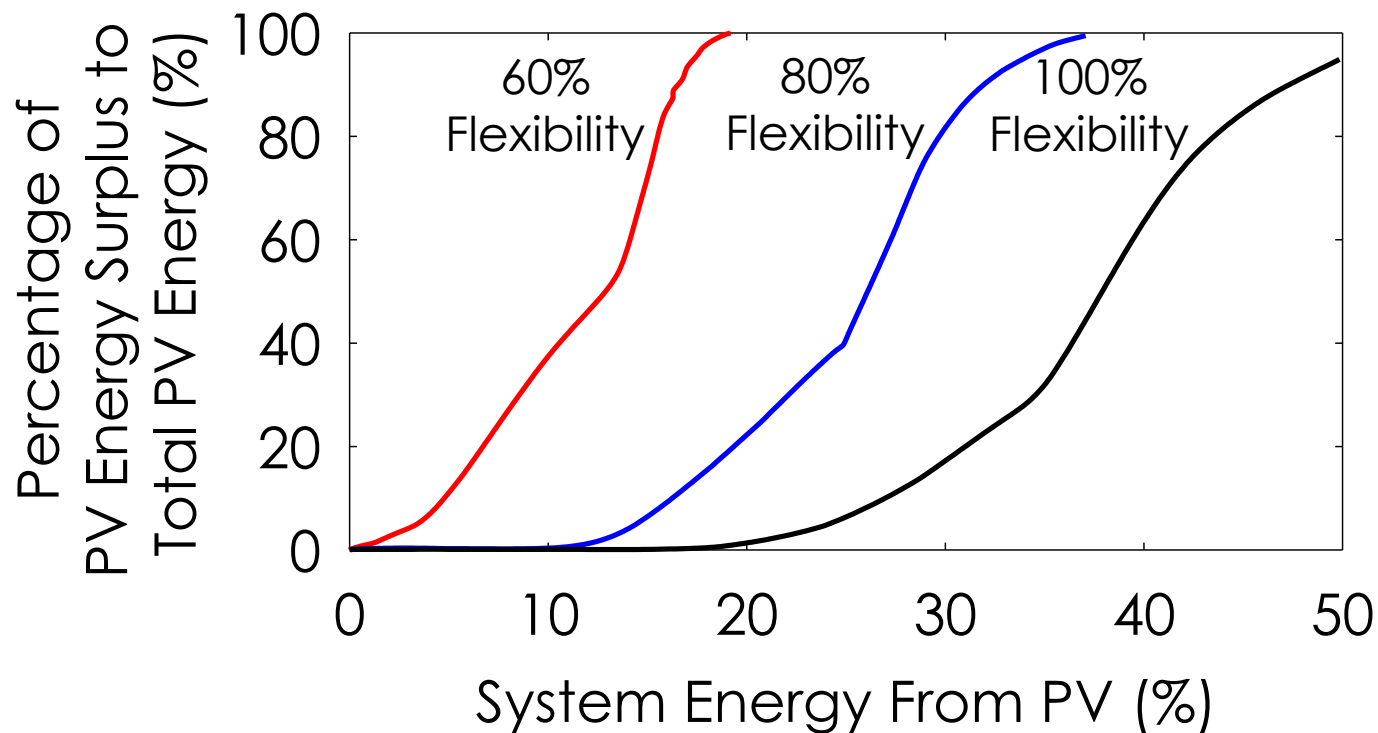
its net energy consumption, by enabling time-shifted matching of internal thermal demand with the diurnal temperature swings of the external natural environment. Second, TES has the ability to make use of inexpensive renewable electricity during its peak production (often oversupply), by storing it in the form most conducive to its final usage — namely as thermal energy for space heating/cooling, instead of electricity.

One fundamental challenge in TES adoption is that there is limited tunability in the usage temperature. For example, if the required temperature is 25 °C and the ambient temperature swings above and below 25 °C, two different TES materials and systems are needed, which dramatically reduces the utilization of each system, leading to a higher cost.

Since the leveled cost of storage (LCOS) is inversely proportional to its utilization

More details in my recent paper:
A. Henry, R. Prasher, A. Majumdar, Nat Energy 5, 635–637 (2020)

The Storage Problem



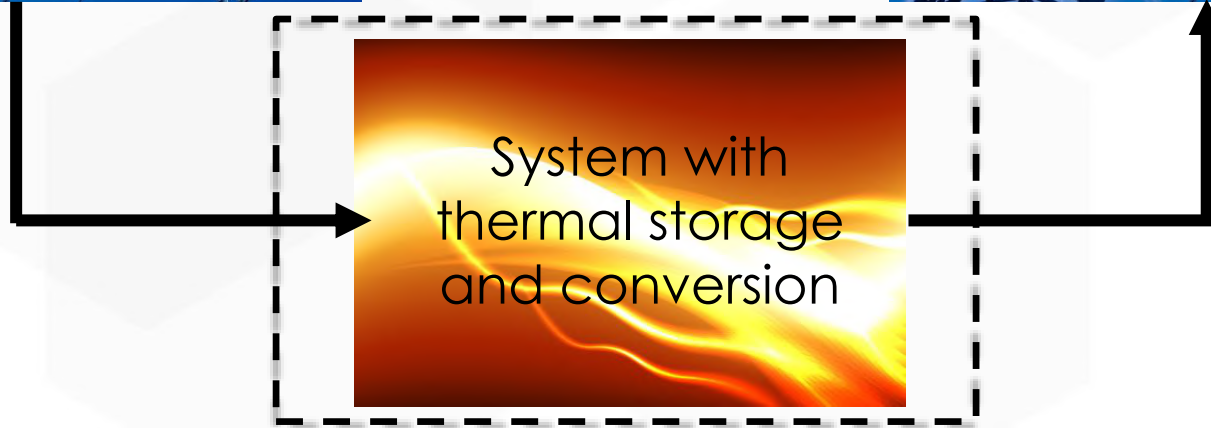
What is Thermal Energy Grid Storage (TEGS)?



Electricity $\rightarrow$ Heat (storage) $\rightarrow$ Electricity



Electricity in
Electricity out
Like a battery



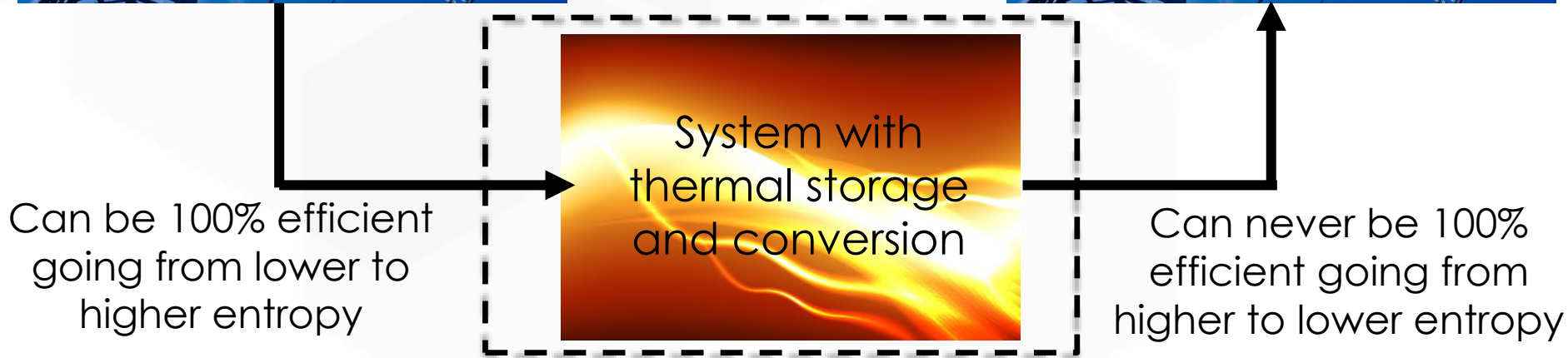
A Thermodynamic Crime!



Electricity $\rightarrow$ Heat (storage) $\rightarrow$ Electricity



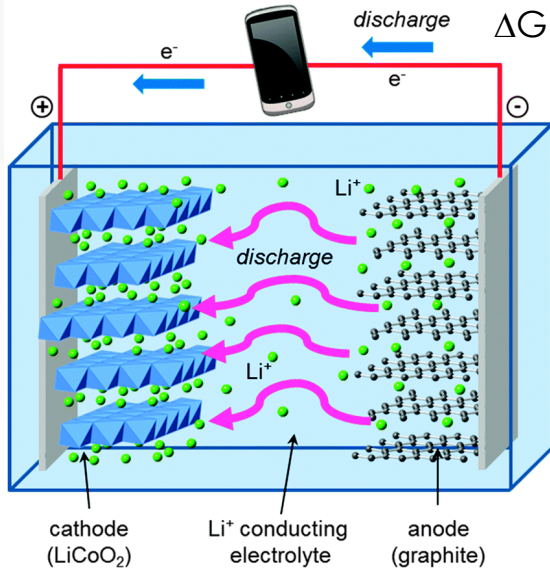
Why would anyone ever do this?
Storing heat can be 10-100X cheaper than storing electricity!



Why is it so cheap? – Atomistic Insight

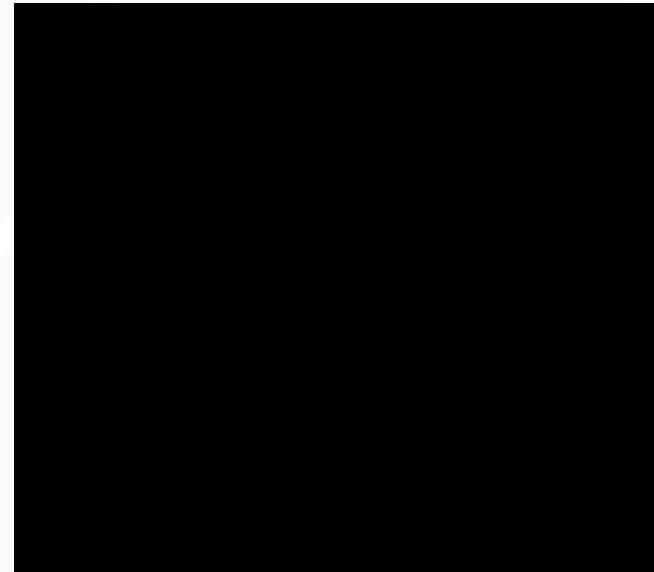


Low concentration of active species
Large energy per active atom
Special, pure, organized materials
Impurities and byproducts are bad



250-695 kWh/m³ **\$150-400/kWh**

100% concentration of active species
Low energy per active atom
Disordered, simple scrap material
Impurity tolerant



300-600 kWh/m³ **\$4-77 kWh**

Simple Estimate



Liquid silicon storage

$$C_p = 950 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$\text{Cost} = \$1.5/\text{kg}$$

$$\Delta T = 500^\circ\text{C}$$

$$\text{Cost/Energy} = \$1.5/\text{kg} \div (C_p \cdot \Delta T) =$$
$$\$11.4/\text{kWh-t}$$

At 50% efficiency

$$\text{Cost/Energy} = \$11.4/\text{kWh} \div 0.5 =$$
$$\mathbf{\$22.8/\text{kWh-e}}$$



Liquid iron storage

$$C_p = 444 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$\text{Cost} = \$0.11/\text{kg}$$

$$\Delta T = 500^\circ\text{C}$$

$$\text{Cost/Energy} = \$0.11/\text{kg} \div (C_p \cdot \Delta T) =$$
$$\$1.8/\text{kWh-t}$$

At 50% efficiency

$$\text{Cost/Energy} = \$1.8/\text{kWh} \div 0.5 =$$
$$\mathbf{\$3.6/\text{kWh-e}}$$

Heat Leakage?



Volume to surface area ratio

$$\tau = m \cdot C_p \cdot R = \rho \cdot V \cdot C_p \cdot L / kA$$

For tanks of order 10 m

τ on the order of months

Loss $\leq 1\%$ of energy stored
per day



What Materials? - Atomistic Insight

What about corrosion?
The hotter the faster/worse!



Si melts at 1414°C

Fe melts at 1538°C

Molten metal dissolves metal
Like
Sugar water dissolves sugar

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H																	2 He
3 Li	4 Be			20 Ca Calcium 40.078 2-8-8-2													10 Ne
11 Na	12 Mg																18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Uut	114 Fl	115 Uup	116 Lv	117 Uus	118 Uuo

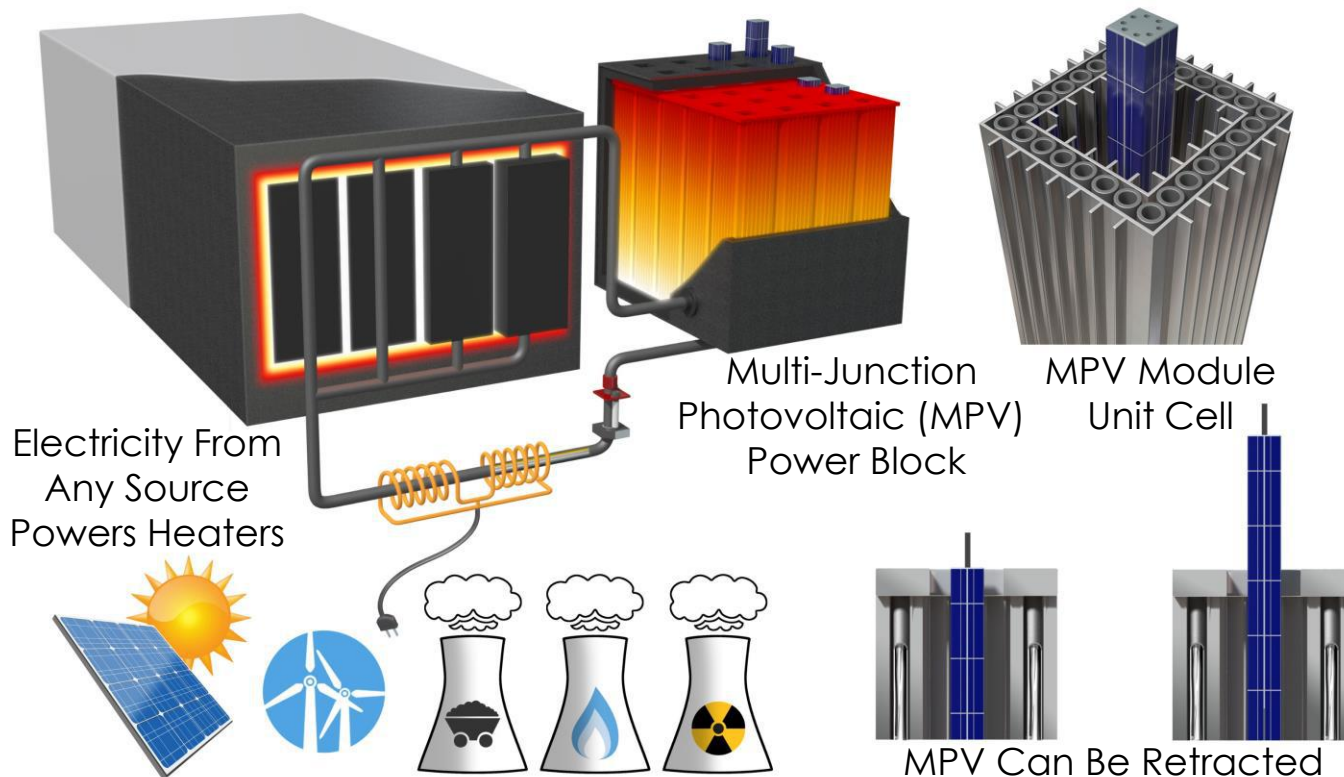
57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

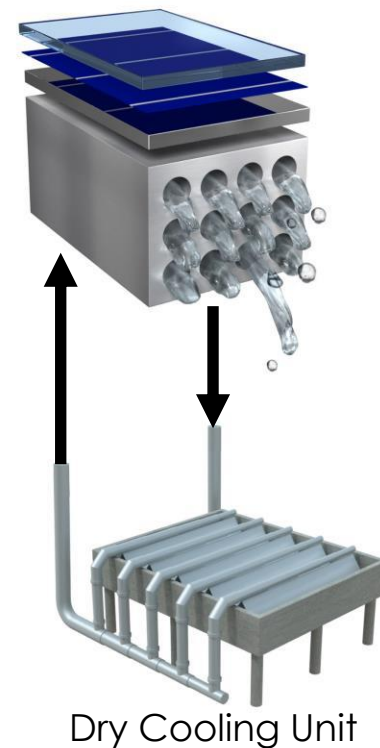
Key New Idea = Liquid Metal + Ceramics

“Sun in a Box” TEGS-MPV

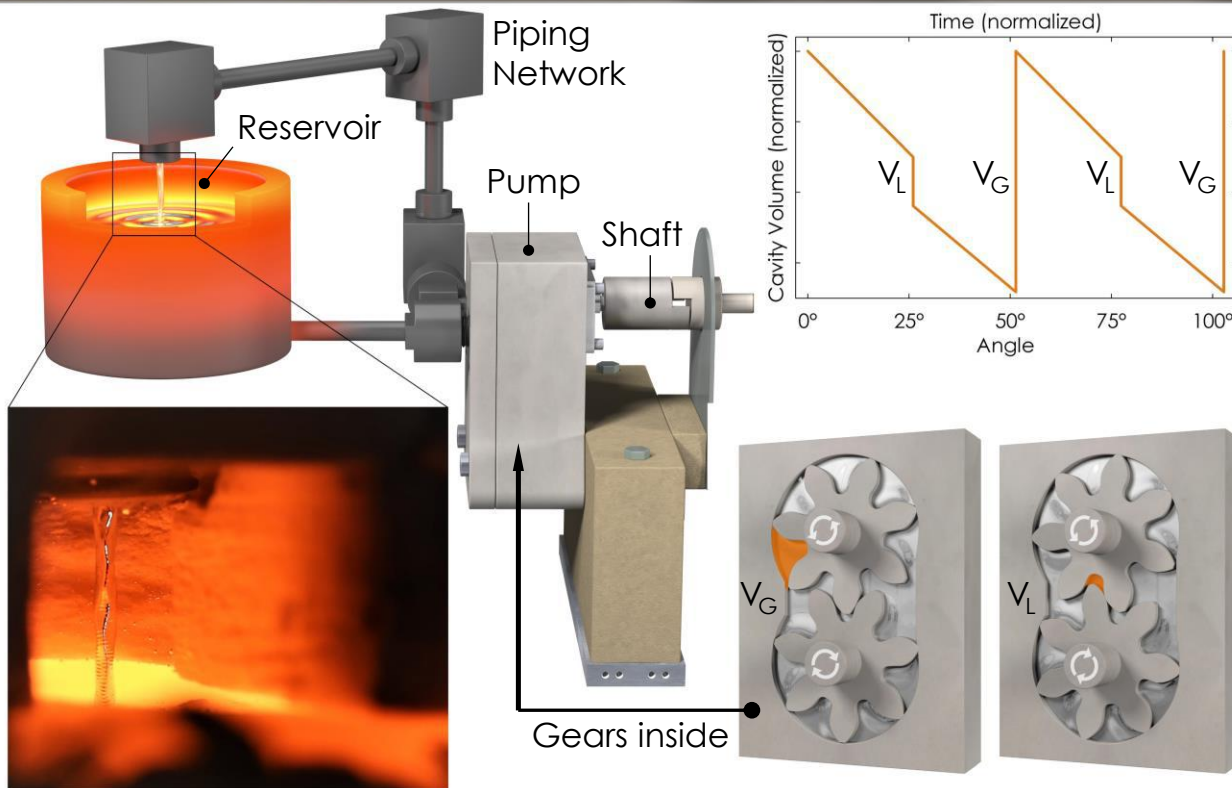
Electricity → Heat → Electricity



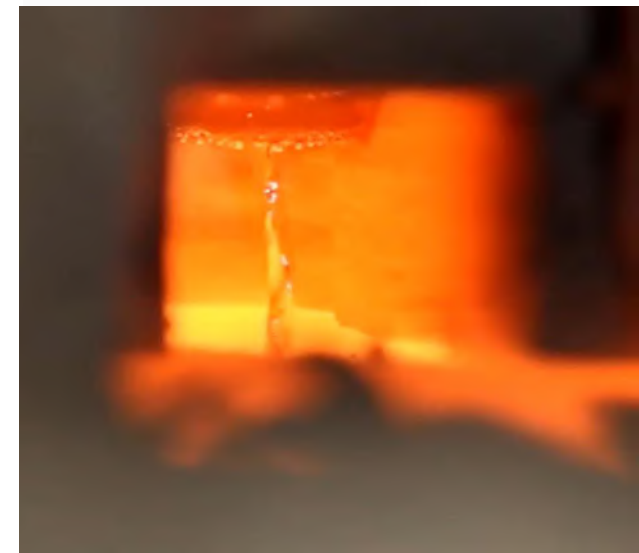
Water Cooled MPV with Integrated Mirror



How Are You Going To Pump It?



Pumping at 1350°C



C. Amy et al., Nature 550, 199–203 (2017)

Why Use MPV Instead of a Turbine?



- Turbine

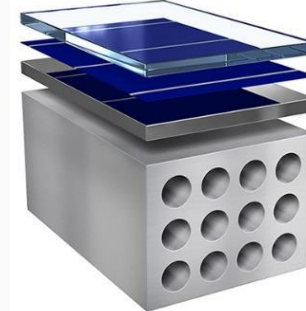
- Doesn't currently exist
- **Large barrier to new turbine deployment**
- > \$100M of R&D
- New materials + New HXs
- Min-Hour response time to full load



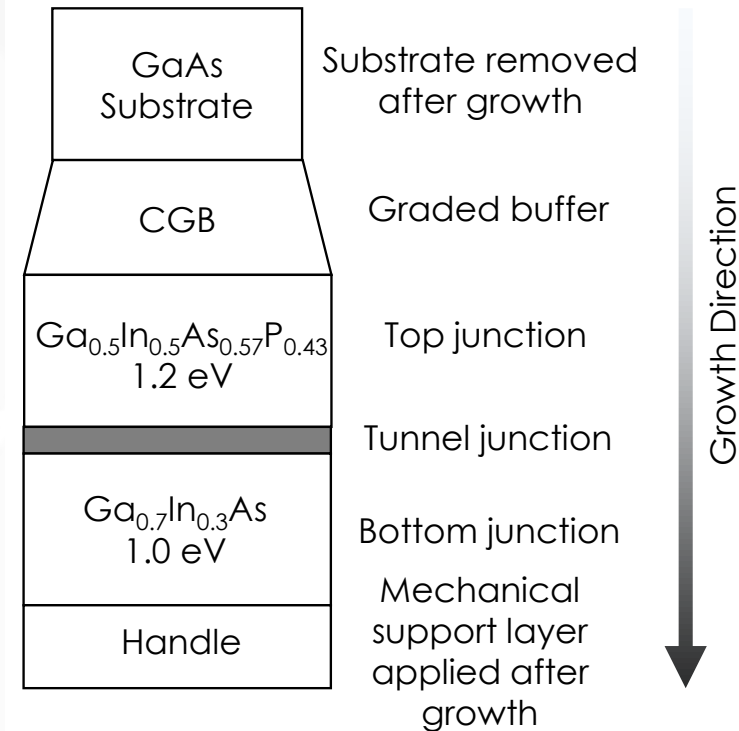
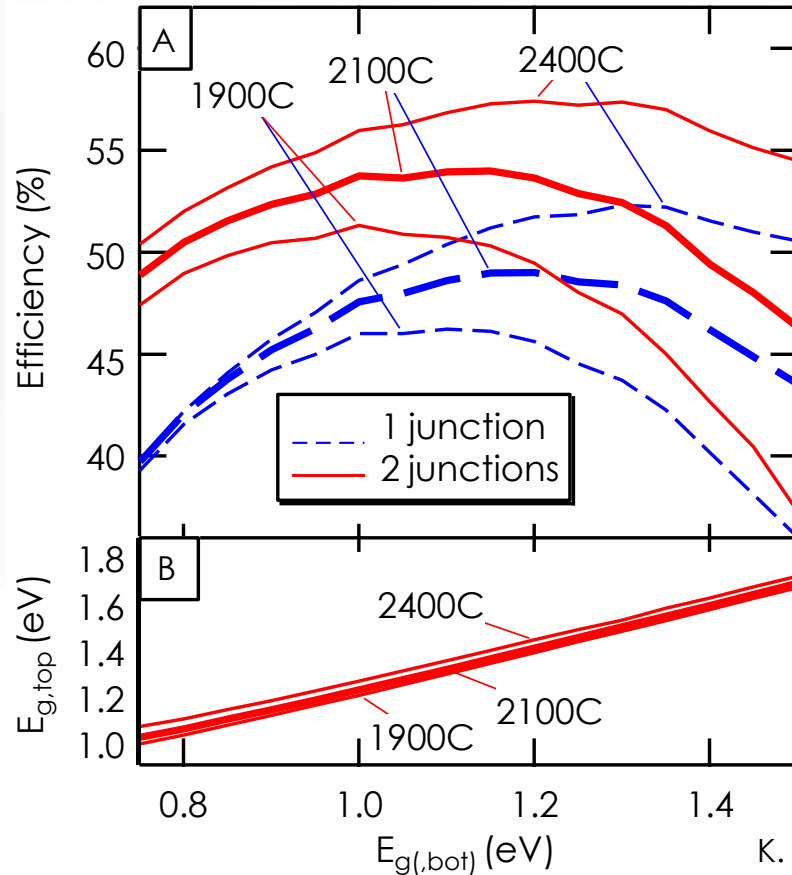
- MPV

- **Much lower barrier to deployment**
- **Lower cost < \$0.5/W-e**
- Similar efficiency (50-55%)
- Fast response time (seconds)
- Fundamentally new cost/learning curve
- Lower maintenance

Turbomachinery
vs.
MPV

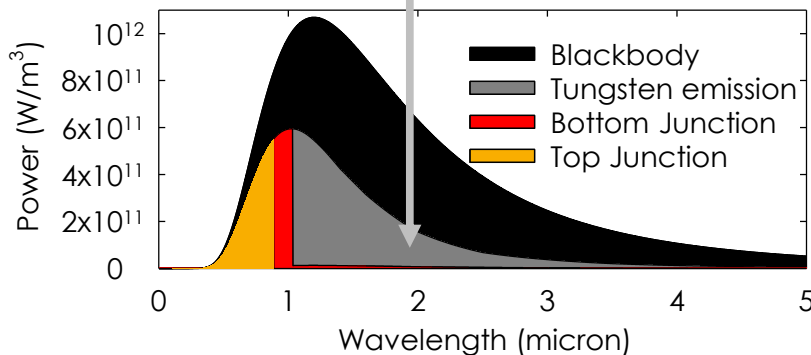


Why Multi-Junction Photovoltaics?



$$Q_{\text{blackbody}} = 1.95 \text{ MW/m}^2$$

Tungsten has low emissivity in the IR



$$Q_{\text{tungsten}} = 689 \text{ kW/m}^2$$

$$Q_{\text{above_BG}} = 213 \text{ kW/m}^2$$

$$Q_{\text{below_BG}} = 476 \text{ kW/m}^2$$

BSR
Reflected
Light

$$\text{Power}_{\text{out}} = 123 \text{ kW/m}^2$$

$$Q_{\text{gen}} = 89.4 \text{ kW/m}^2$$

$$Q_{\text{below_BG}} = 18.7 \text{ kW/m}^2$$

Black Body Incident Light Absorbed Light

$$\text{Efficiency} = \text{Power}_{\text{out}} \div Q_{\text{total}} = 123 \div (123 + 89.4 + 18.7 + 4.6) = \mathbf{52\%}$$

- <1% loss in electronics for heater
- ~1%/day loss in heat leakage
- <1% loss in parasitic load
- ~ 50% roundtrip efficiency (RTE)

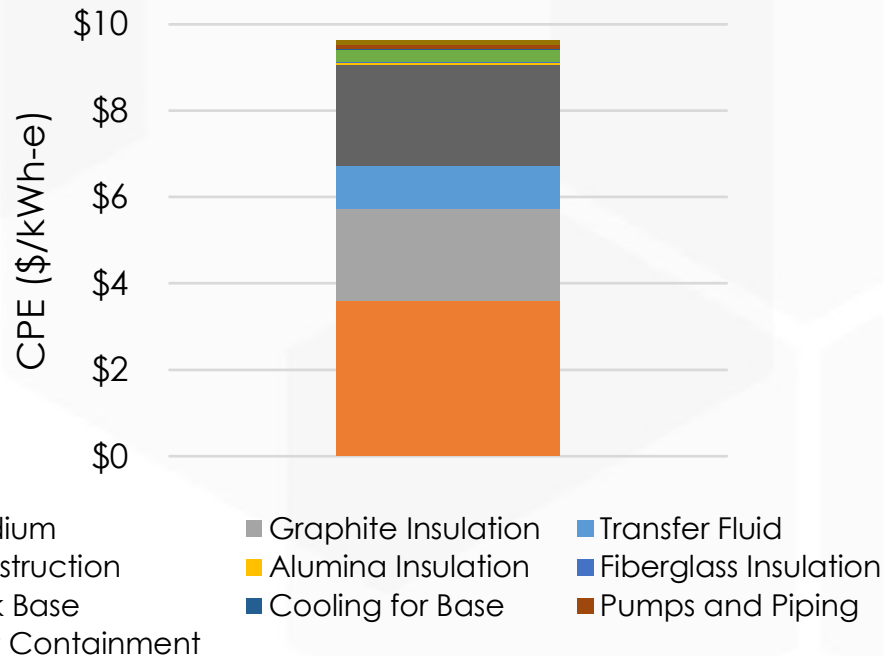
C. Amy et al., Energy & Environmental Science, 12, 334 (2019)

System Cost

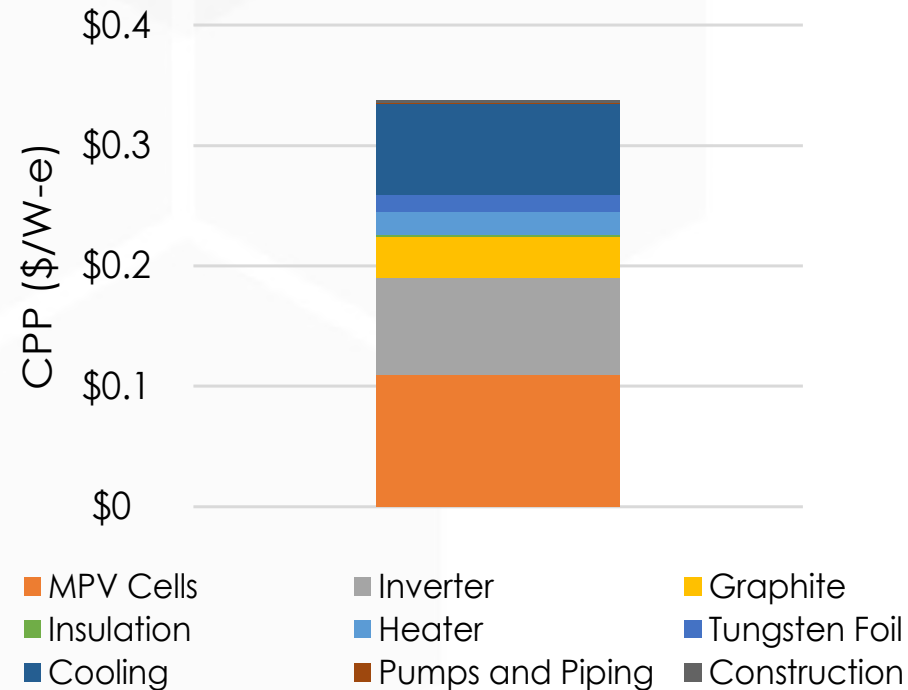


$$\text{Cost} = \text{CPE} \times \text{time} + \text{CPP}$$

Cost per unit energy = CPE



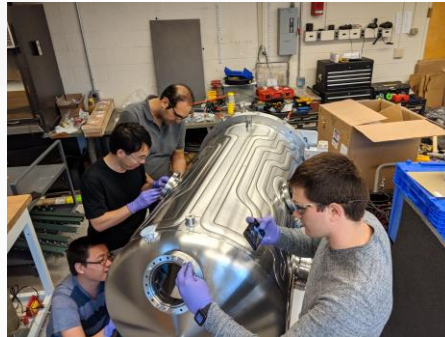
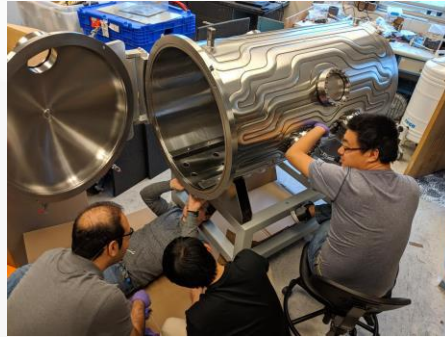
Cost per unit power = CPP



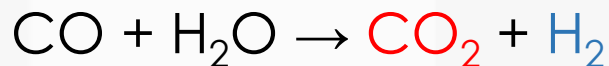
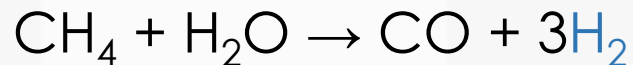
What's Next?



- ARPA-E Project
- **Build a prototype**
- Pumping
- 2500°C Heaters
- Emitter evaporation/deposition
- Cell redesign/optimization + fabrication
- High current density
- High reflectivity ($> 98\%$)
- High efficiency ($\geq 50\%$)
- Long term testing

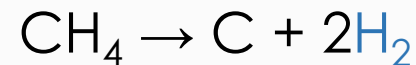


All Gases



Steam Methane Reforming (SMR)
+
Water-Gas Shift (WGS)

Gas + Solid Products



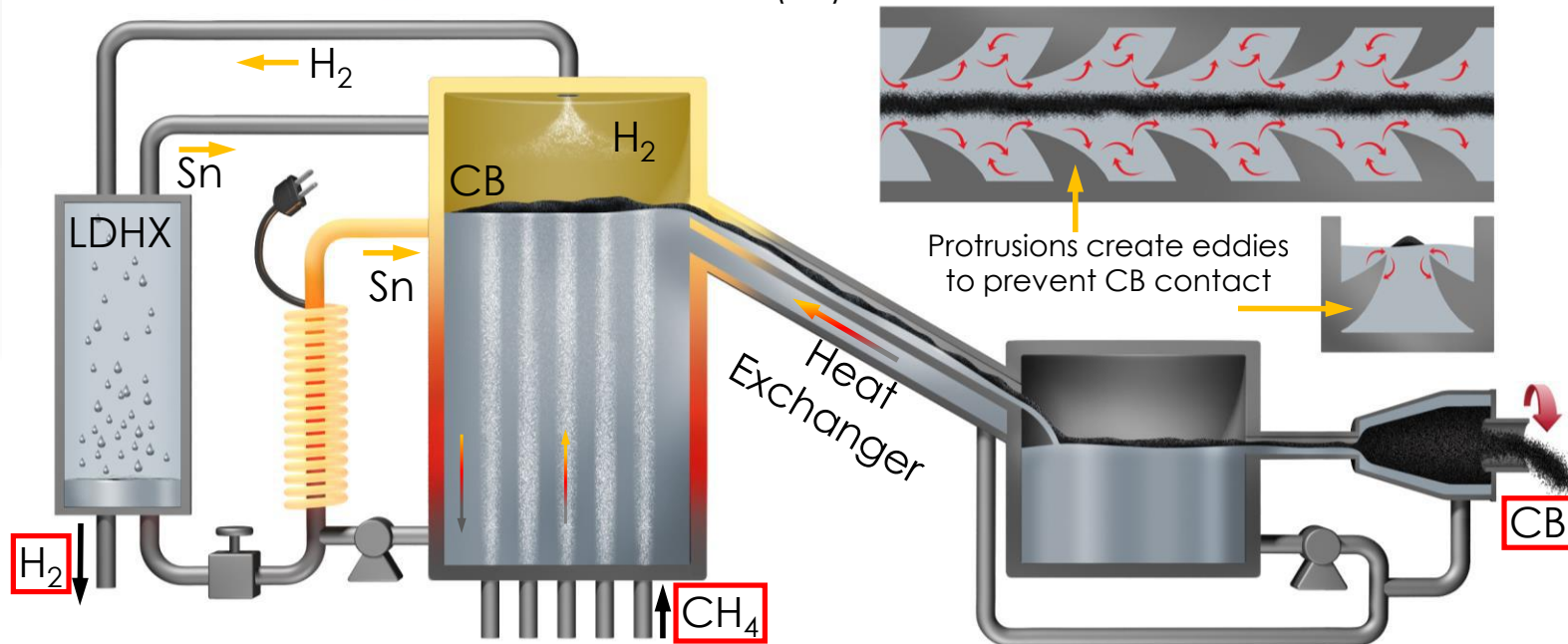
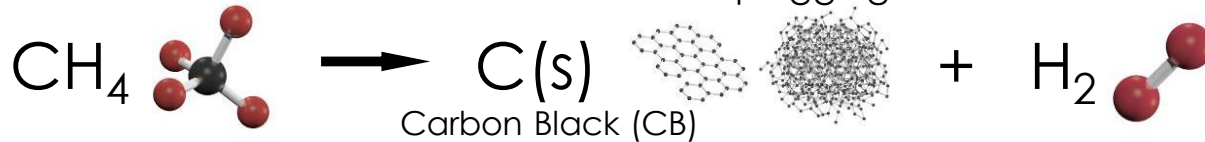
Direct Thermal Pyrolysis



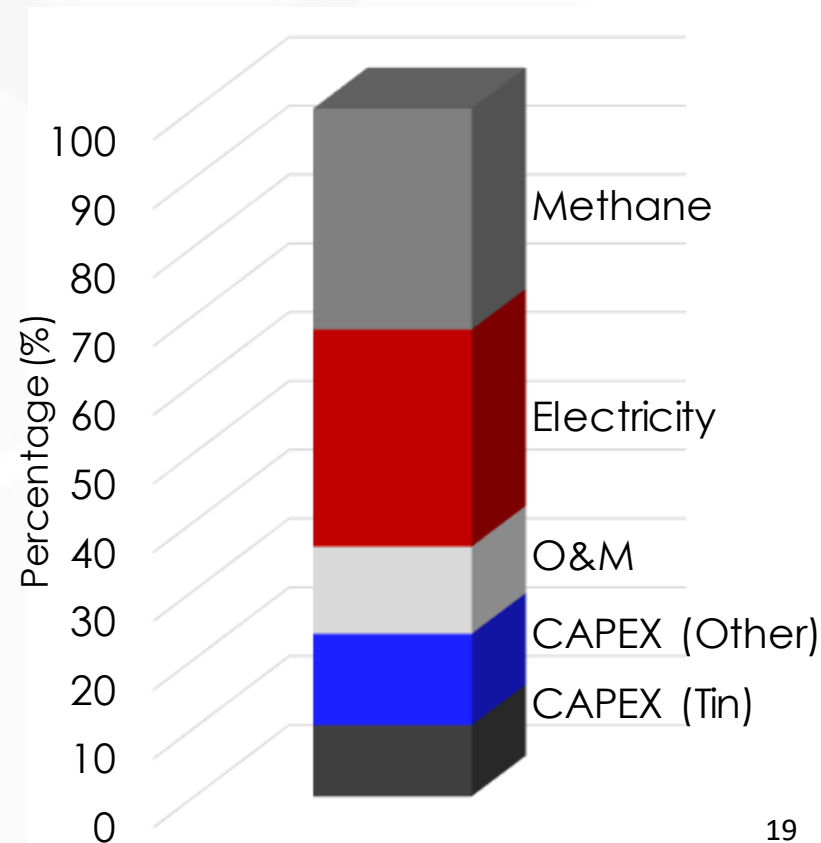
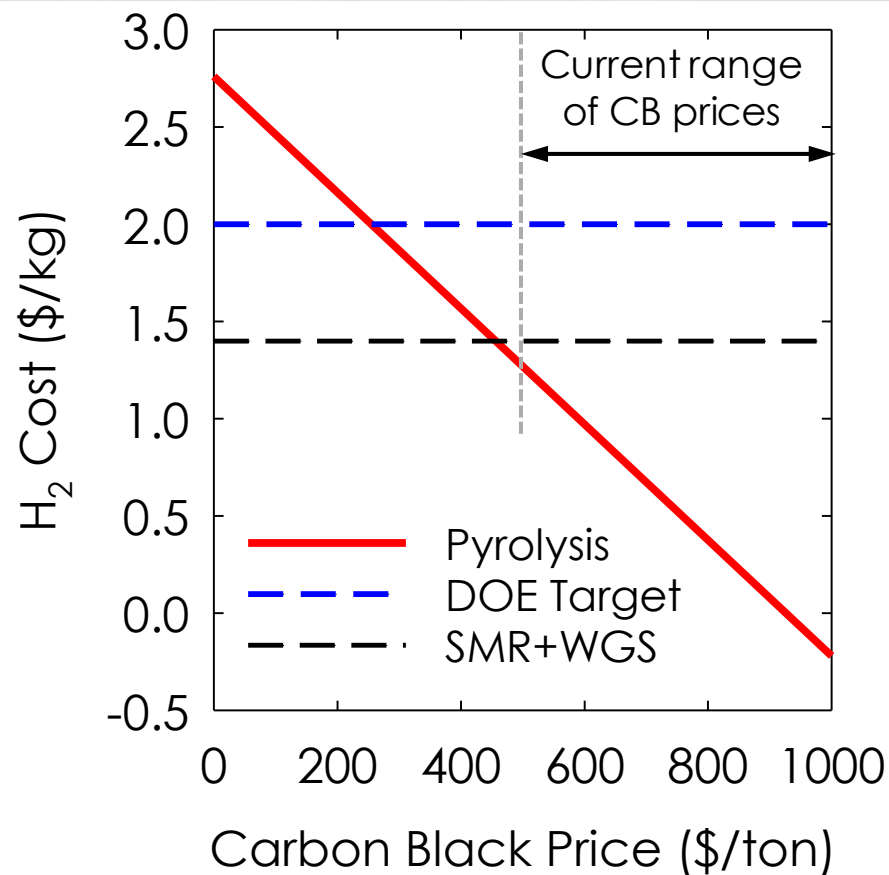
The plugging problem

THE REACTOR CONCEPT

- Lower cost H_2 \$0.5-1.50/kg
- No CO_2 emissions
- Produces carbon black
- No corrosion – No plugging!
- We now know how to make it
- We know how to pump $Sn(l)$

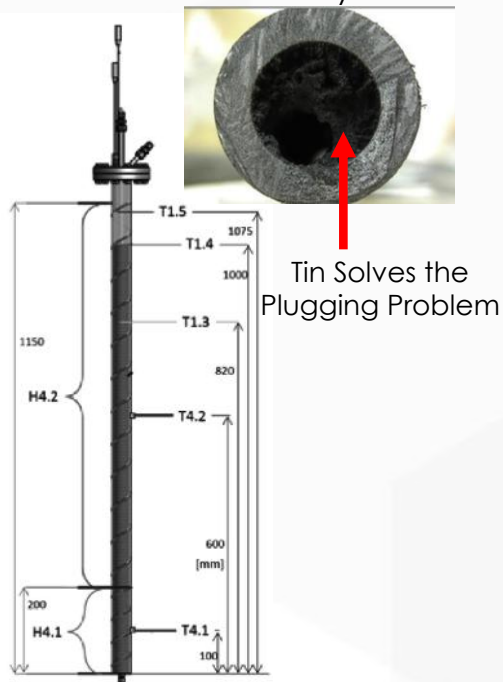


THE COMPELLING ECONOMICS



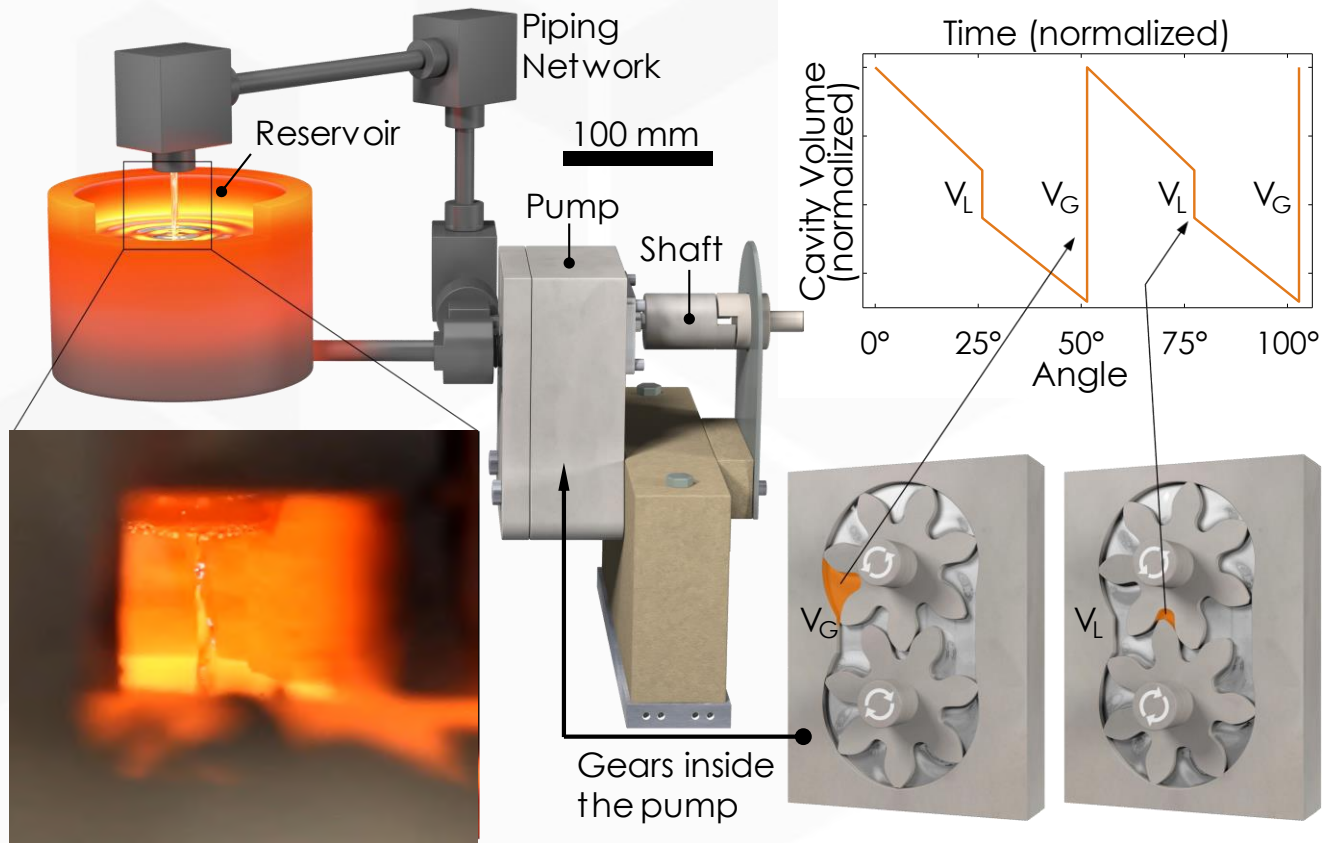
PRIOR WORK – LAYING THE FOUNDATION

Wetzel's Group
at KIT in Germany



M. Plevan *et al.* Int. J Hydrogen
Energy, 40, 25, 8020-8033 (2015)

The PI's Group - C. Amy *et al.*, Nature, 550,199–203 (2017)



PRIOR WORK – LAYING THE FOUNDATION

For $T \leq 500^{\circ}\text{C}$

Union Joints



Elbows



Pumps



Valves

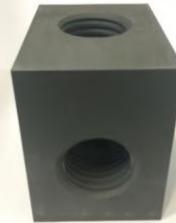


For $T \geq 500^{\circ}\text{C}$

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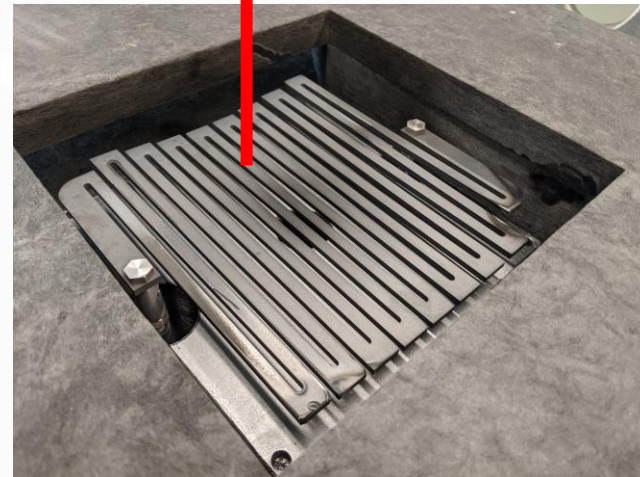
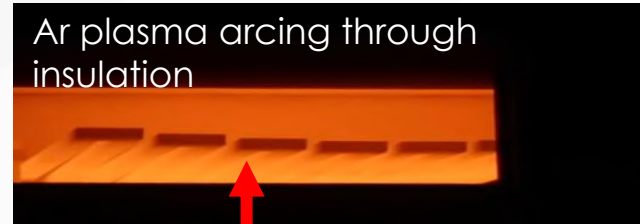


Valves



> 2000°C Heaters

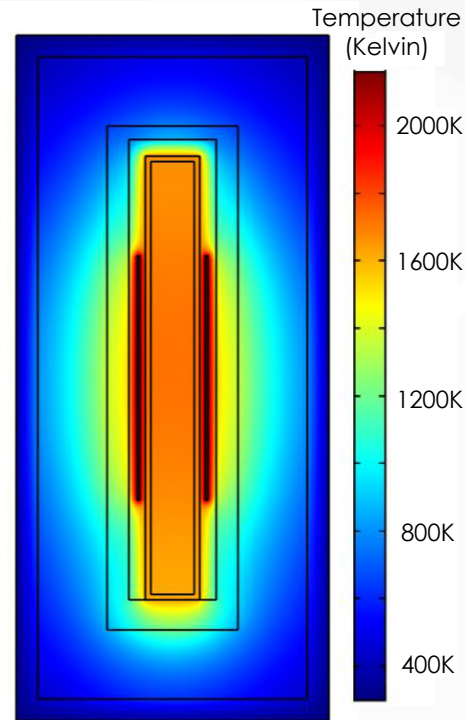
Ar plasma arcing through insulation



TASK 1: REACTOR DEVELOPMENT

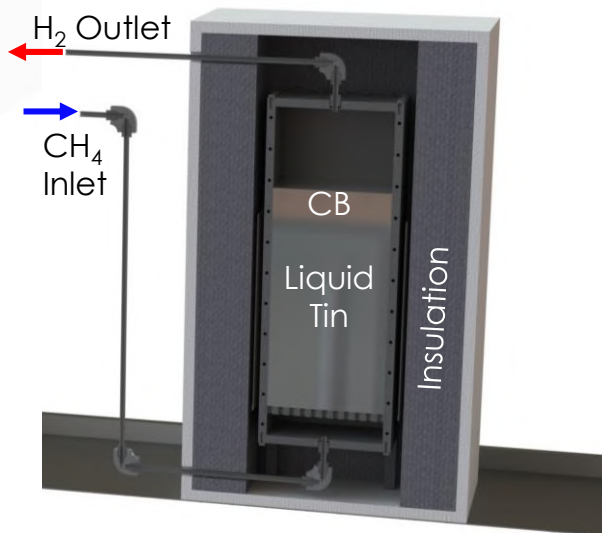


COMSOL Modelling Guided Experimental Design

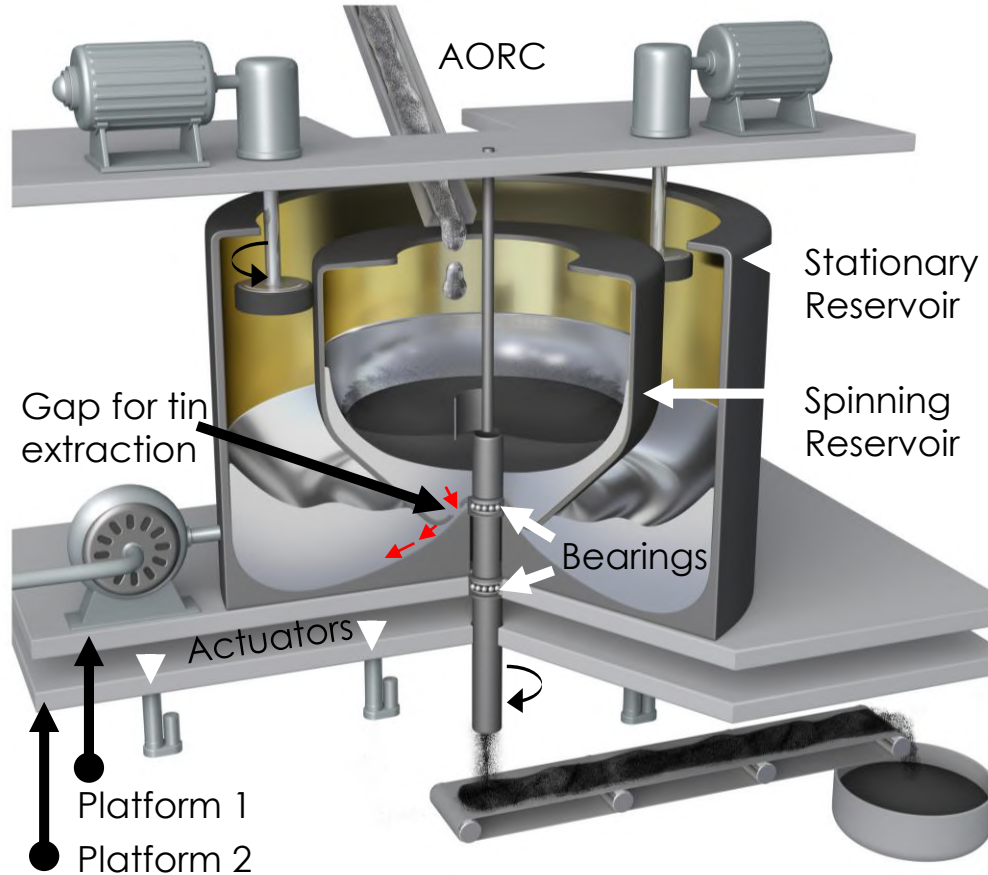


Detailed reactor modeling:

- Predict profiles of temperature, species concentration and bubble size density in reactor
- PDAEs in time, vertical position & bubble size coordinate
- Software: Jacobian (equation-based modeling system) & DAEPACK (numerical engine)
- Incorporate detailed reaction kinetics models for methane pyrolysis via, e.g., CHEMKIN
- Optimization-based experimental design with reactor model for model discrimination and parameter estimation
- Cycle of design, experiments, validation

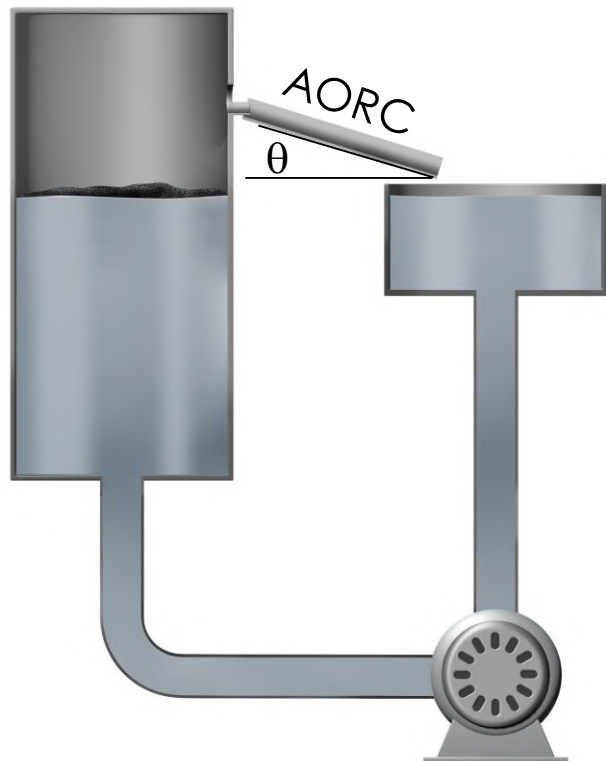


TASK 2: CENTRIFUGE DEVELOPMENT

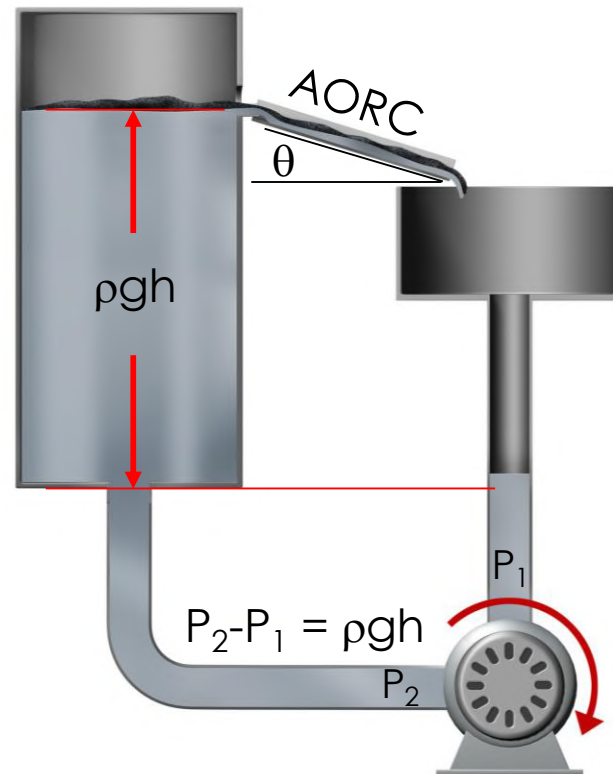


TASK 3: CONTINUOUS REACTOR SYSTEM DEVELOPMENT

Before Pumping



While Pumping

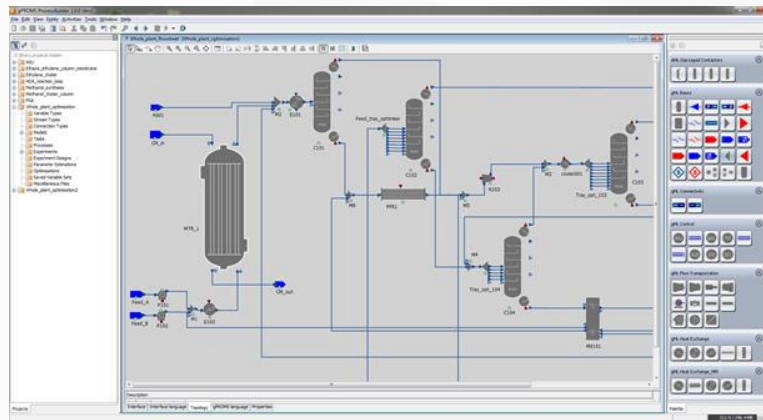


TASK 4: INDUSTRIAL SCALE SYSTEM MODELING

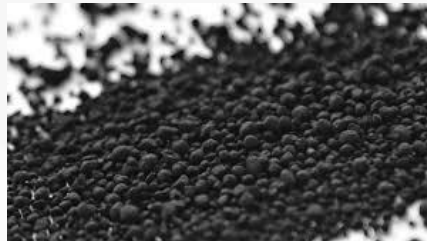
Liquid Droplet
Heat Exchanger
Model



Full System Model for Performance



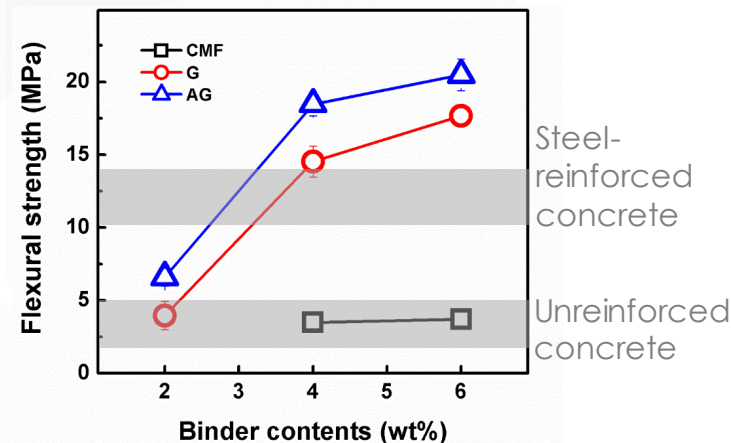
Carbon Black (CB)



Hydrogen



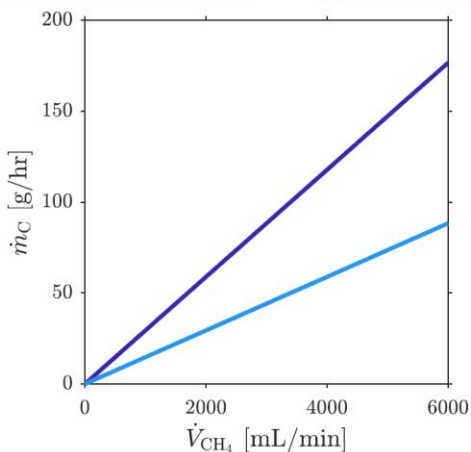
Yu Qiao (UCSD)
Construction Material



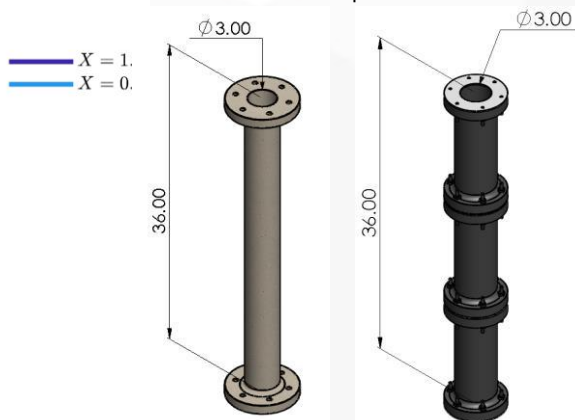
Technoeconomics



PROGRESS TO DATE



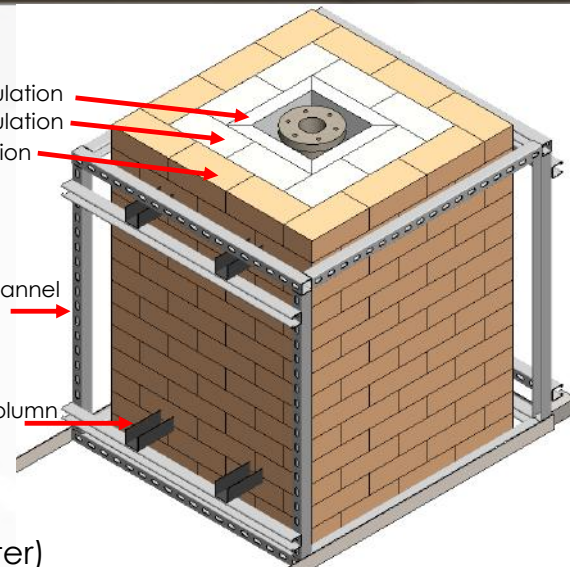
SiC vs. Graphite Reactor?



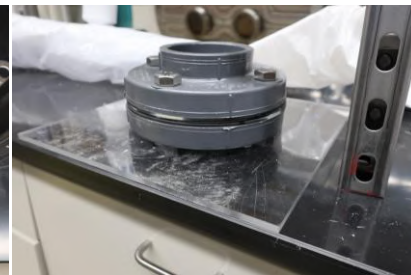
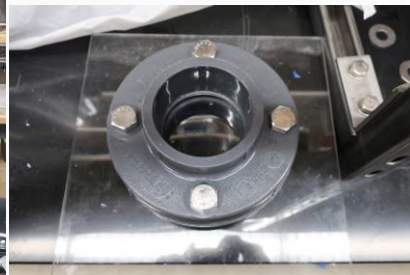
Zirconia-based insulation
 Alumina-based insulation
 Silica-based insulation

Stainless steel strut channel

CFC support for column



Mock reactor (water)



Five thermal energy grand challenges for decarbonization

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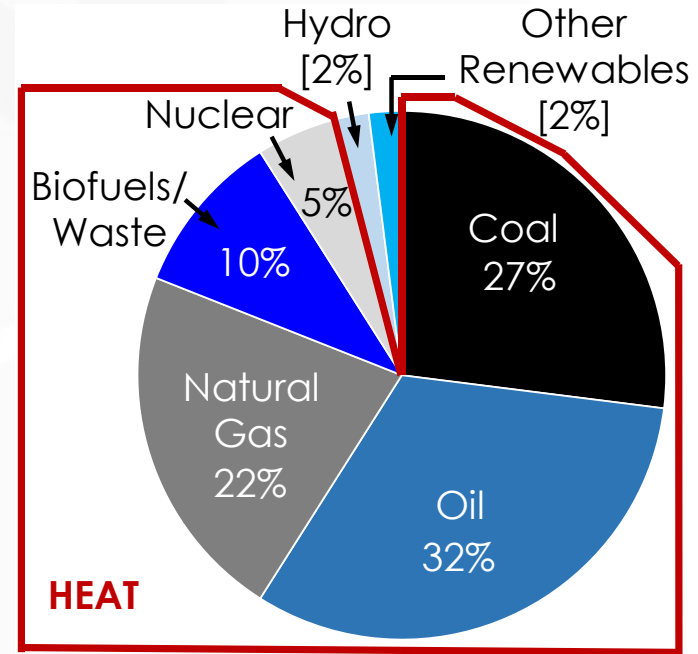
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Since the levelized cost of storage (LCOS) is inversely proportional to its utilization

nature
energy

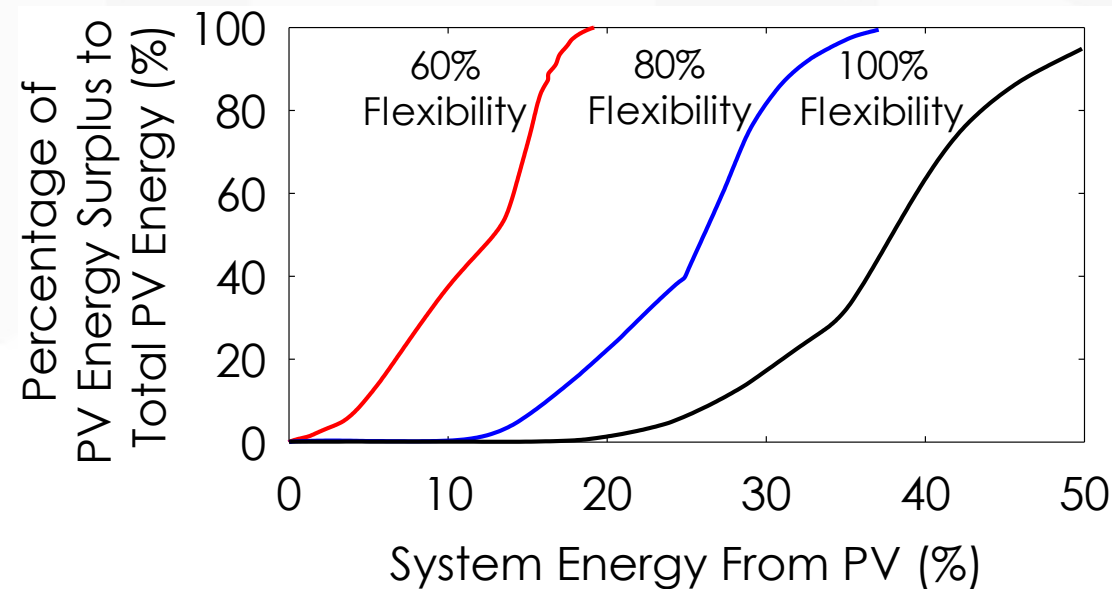


Global primary energy usage in 2016
– IEA Energy Outlook

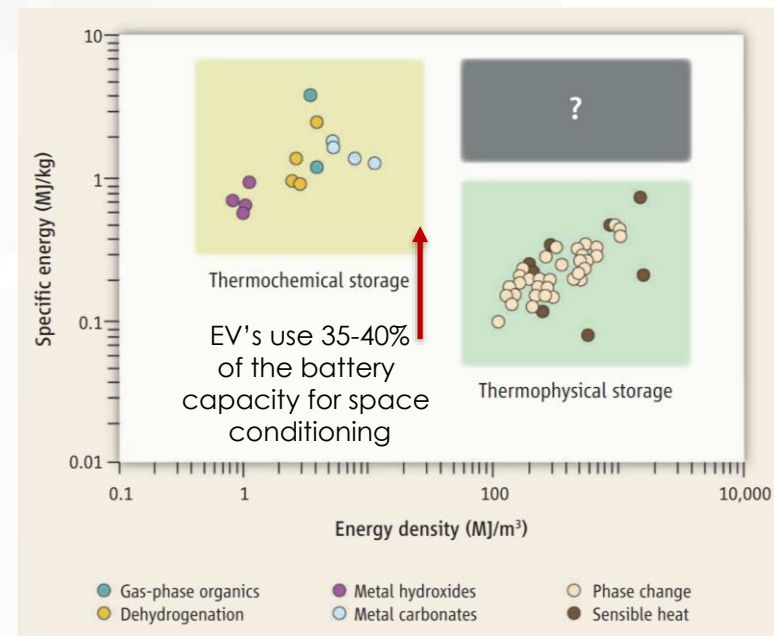
High Energy Density Thermal Storage



P. Denholm, R. Margolis, *Energy Policy*, 35, 2852–2861 (2007)



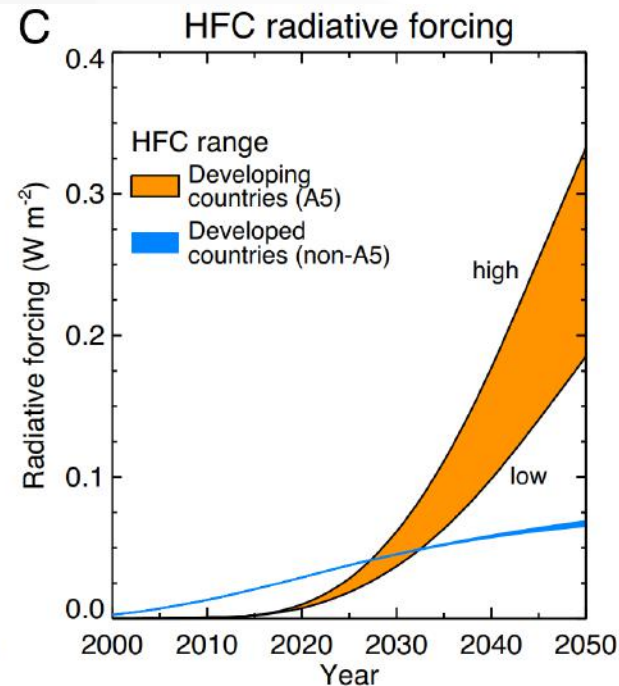
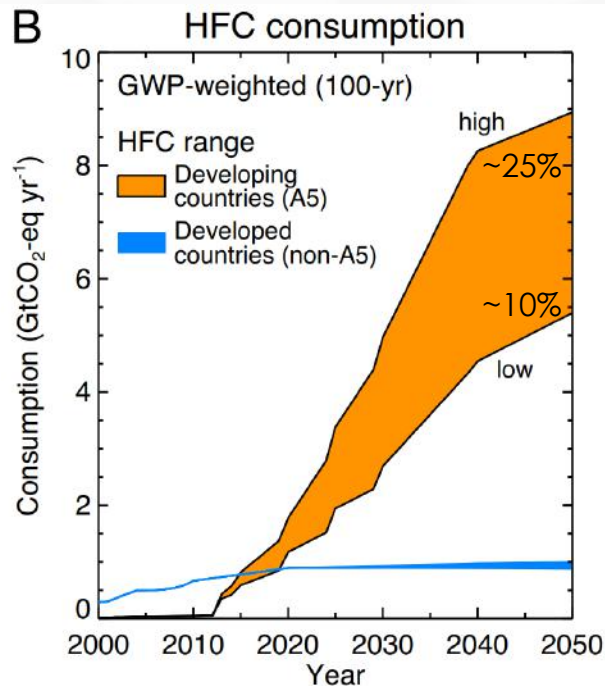
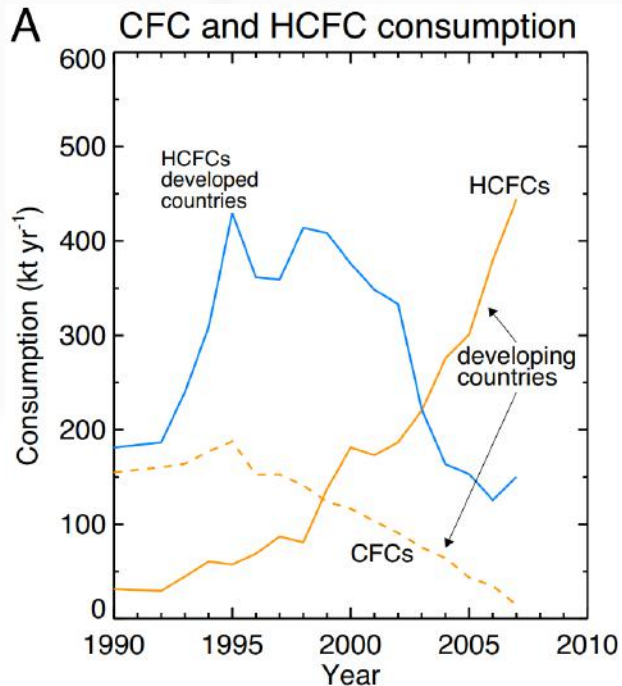
I. Gur et al. *Science*, 435, 1454 (2012)



Reinventing Cooling



Global CO₂ emissions ~ 37 GtCO₂-eq → HFCs could become 10-25% of the problem!

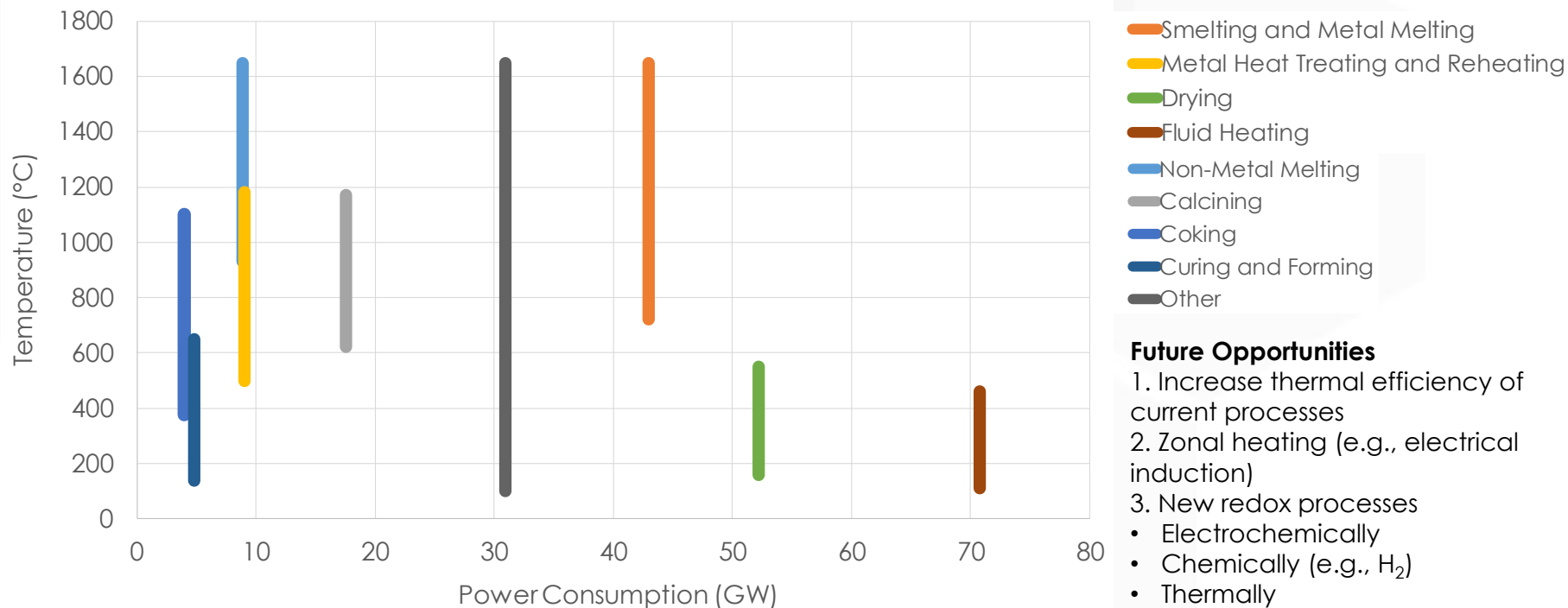


G. Velders et al. PNAS, 106, 27, 10949-10954 (2009)

Decarbonizing the Industrial Sector



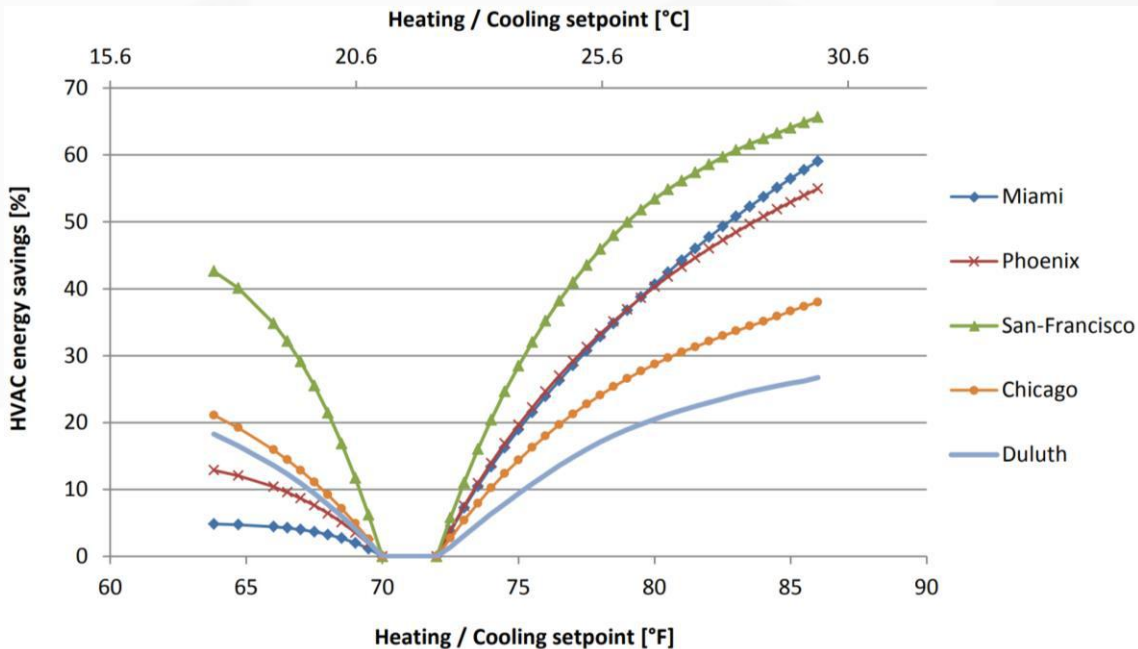
Electricity: 25% Transportation: 14% **Industry: 15%** → Cement: 5% Steel: 4% Aluminum: 1% Hydrogen: 1%



Zonal Heating and Cooling



Space heating and cooling ~ 13% of US energy usage

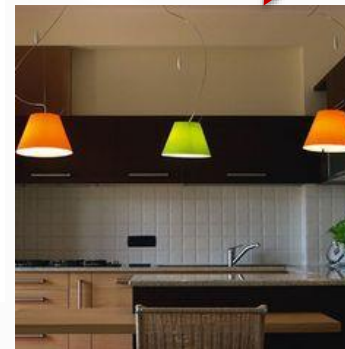


T. Hoyt et al., Building and Environment, doi:10.1016/j.buildenv.2014.09.010

Building



Local



Wearable

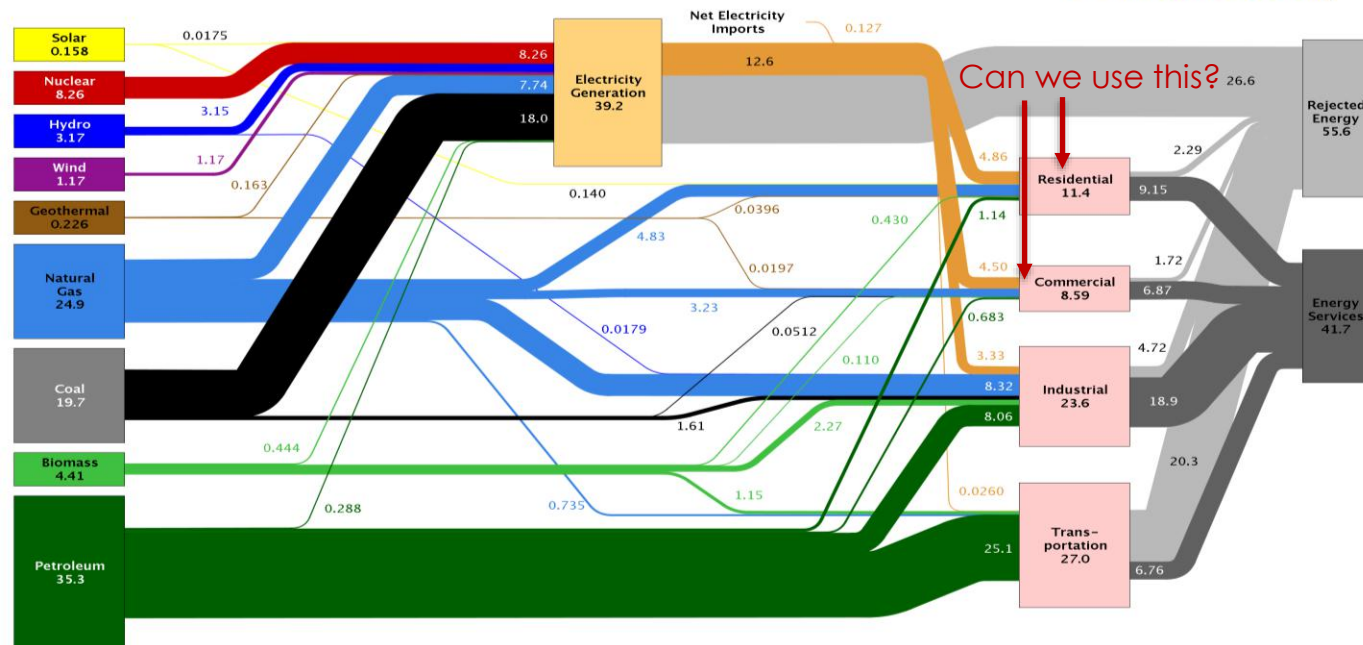


Long Distance Transmission of Heat

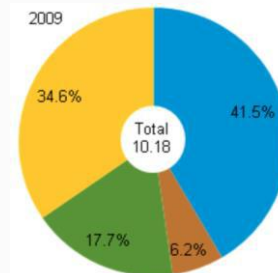
building energy goes to heating at $< 60^{\circ}\text{C}$
 ~ 60% for residential
 ~ 32% for commercial

Estimated U.S. Energy Use in 2011: ~97.3 Quads

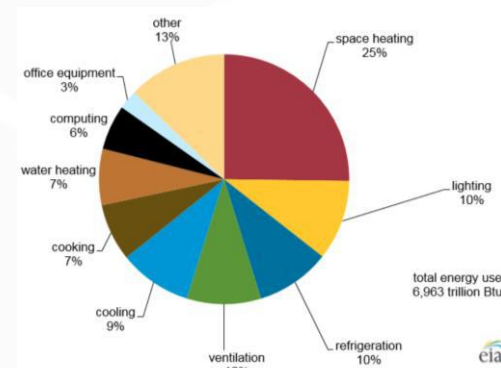
Lawrence Livermore
National Laboratory



Residential



Commercial



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Selected Collaborators:

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- Paul Barton (MIT)
- Samuel Graham (GT)

Former Group Members:

- Dr. Diane England
- Dr. Cansheng Yuan
- Dr. Caleb Amy
- Dr. Wei Lv
- Dr. Hamid Seyf
- Dr. Freddy DeAngelis
- Dr. Murali Muraleedharan
- Dr. Xu Tan
- Dr. Bamdad Barari
- Colby Jarrett
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- Malavika Bagepalli
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QUESTIONS?