

# A brief glimpse at a role for data science in GHG mitigation pathways, materials focus

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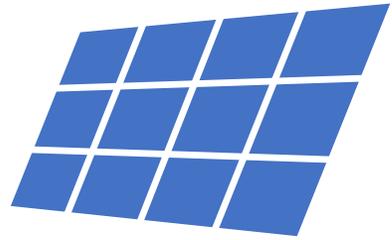


**MATERIALS  
PROJECT**



**MITe<sup>i</sup>**  
MIT Energy Initiative

# Scale up time measured in decades



Bell  
Labs

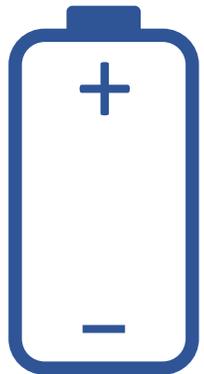
1954

Sharp

1963

Kyocera

1982



1980

Oxford

1991

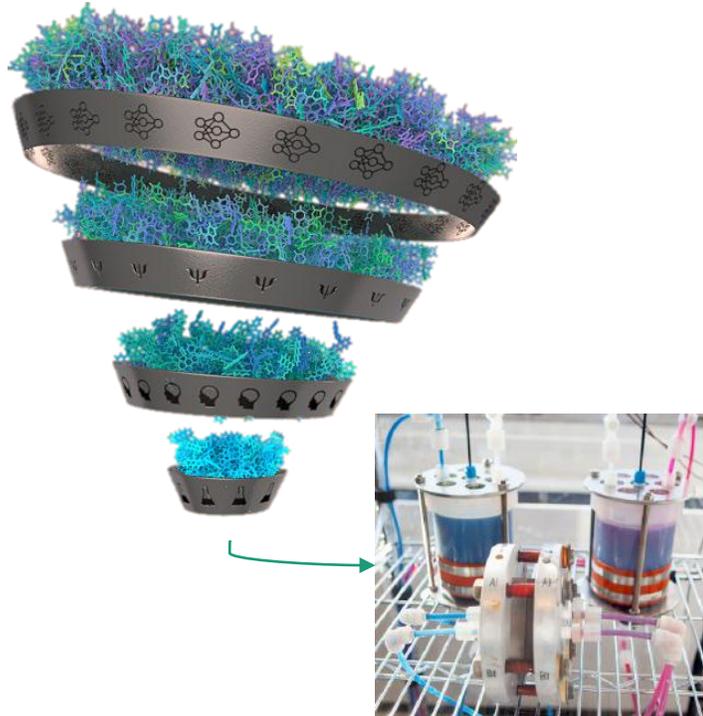
Sony

2010

Chevrolet



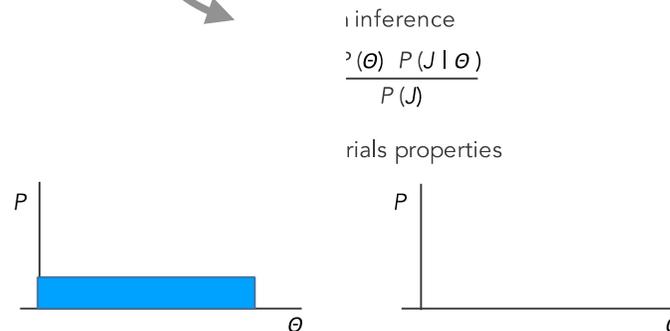
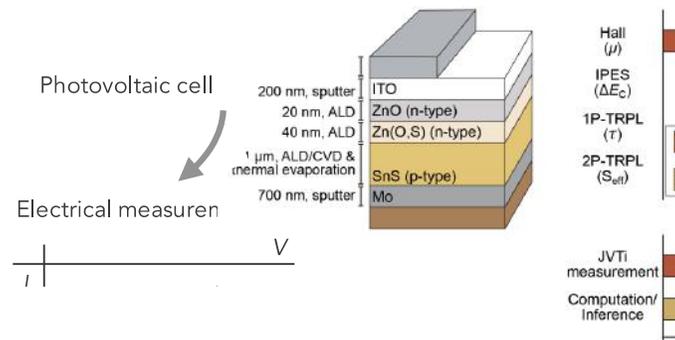
# Vision for computer aided, data-driven materials development



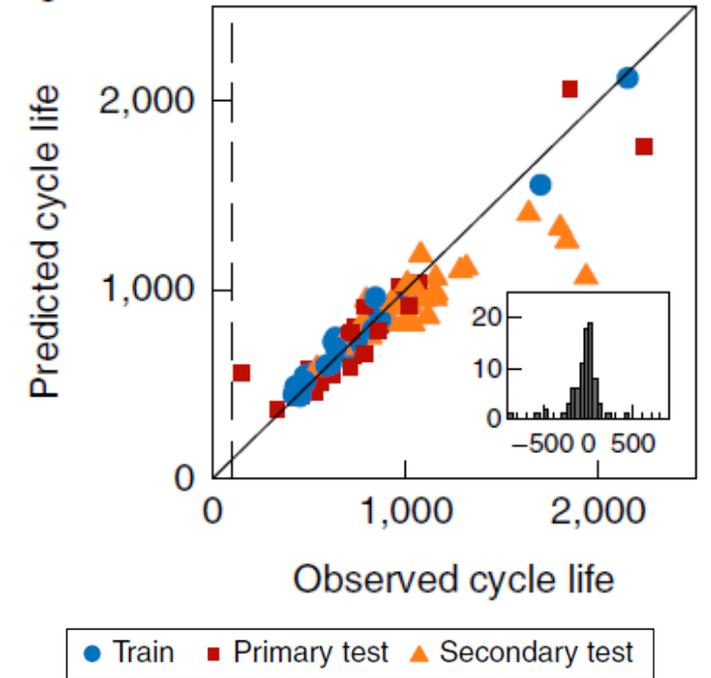
High-throughput virtual screening for promising materials

G-B. et al. *Nature Materials* 2016

## Device optimization and high-throughput experimentation

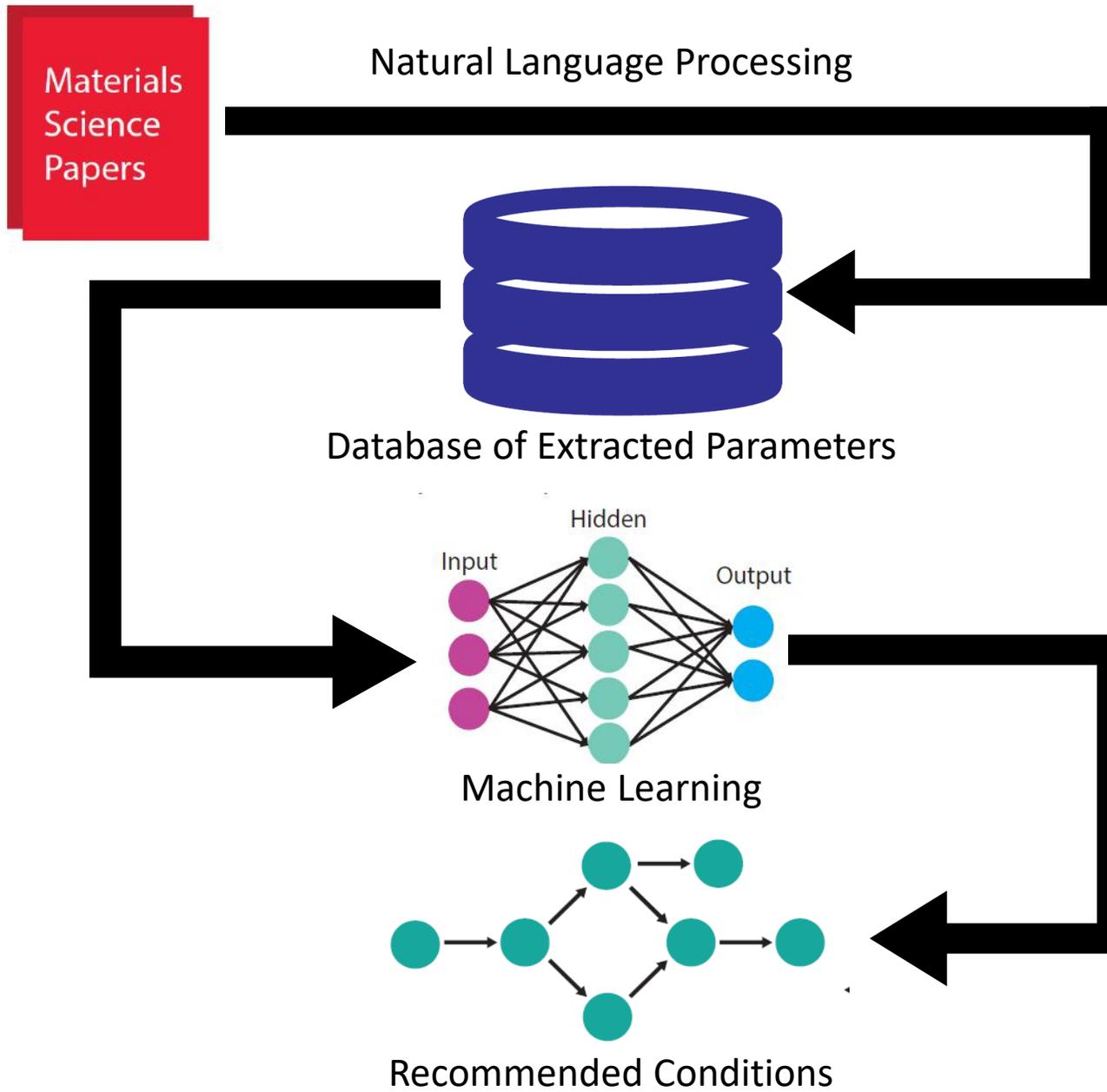


R.E. Brandt et al. *Joule* 2017



Predicting lifetime of complex, nonlinear systems such as lithium-ion batteries

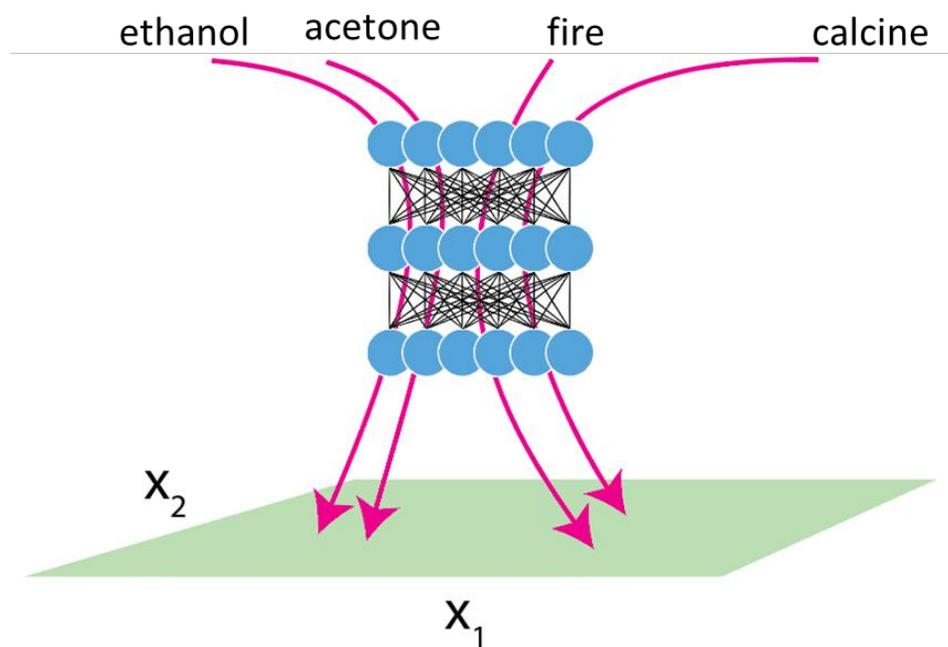
Severson et al. *Nature Energy* 2019



# This is a semi-supervised problem

## Unsupervised word embeddings

Projections of words into real-value vectors where synonyms are projected to nearby coordinates



## Supervised classification

Electrochemistry Communications 38 (2014) 79–81

Contents lists available at ScienceDirect

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Short communication

Electrochemical properties of  $\text{NaNi}_{1/3}\text{Co}_{1/3}\text{Fe}_{1/3}\text{O}_2$  as a cathode material for Na-ion batteries

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Department of Materials Science and Engineering, Massachusetts Institute of Technology, Cambridge, MA 02139, USA

**1. Introduction**

Na-intercalation batteries are appearing as an important alternative to Li intercalation systems, and rapid progress has been made on developing high capacity cathode materials [1]. It has become clear that the Na analogues of the successful layered  $\text{LiMO}_2$  electrodes behave very differently from their Li equivalents [2]. The large difference in ionic radius between Li and Na provides a stronger tendency for the Na compounds to form in the layered structure [3–7], and layered  $\text{Na}_x\text{MO}_2$  ( $M = \text{Ti, V, Cr, Mn, Fe, Co, Ni}$ ) [4,8–17], as well as several Na compounds with mixed transition metals,  $\text{Na}_x\text{Ni}_{0.6}\text{Co}_{0.4}\text{O}_2$ ,  $\text{Na}_x\text{Ni}_y\text{Mn}_{1-y}\text{O}_2$ ,  $\text{Na}_x\text{Ti}_y\text{Mn}_{1-y}\text{O}_2$ ,  $\text{Na}_x\text{Fe}_{0.5}\text{Mn}_{0.5}\text{O}_2$ ,  $\text{NaFe}_{0.5}\text{Co}_{0.5}\text{O}_2$ ,  $\text{NaNi}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3}\text{O}_2$  and  $\text{NaNi}_{1/3}\text{Fe}_{1/3}\text{Mn}_{1/3}\text{O}_2$  [18–27], all show electrochemical activity.

In this paper, we report the synthesis and electrochemical performance of  $\text{NaNi}_{1/3}\text{Fe}_{1/3}\text{Co}_{1/3}\text{O}_2$  as a novel Na intercalation cathode material. The only layered materials in which three transition metals are mixed in the literature are  $\text{NaNi}_{1/3}\text{Fe}_{1/3}\text{Mn}_{1/3}\text{O}_2$  [26] and  $\text{NaNi}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3}\text{O}_2$  [24], and their capacity is limited to about

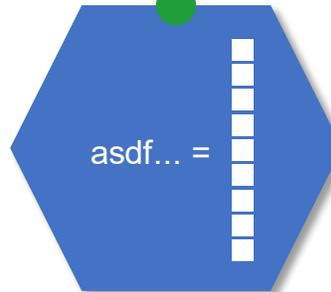
material is de-sodiated the Fe and Co are first oxidized to 4+ while the Ni goes through two oxidation steps to reach a final oxidation state in the fully de-sodiated structure of  $\text{Co}^{4+}$ ,  $\text{Fe}^{4+}$  and  $\text{Ni}^{4+}$ .

**2. Experimental methods**

$\text{NaNi}_{1/3}\text{Co}_{1/3}\text{Fe}_{1/3}\text{O}_2$  was synthesized by solid-state reaction. Excess amounts of  $\text{Na}_2\text{O}$ ,  $\text{NiO}$ ,  $\text{Co}_3\text{O}_4$  and  $\text{Fe}_2\text{O}_3$  were mixed and ball milled for 4 h at 500 rpm rate, and the resulting material was collected in the glove box. About 0.5 g of powder was fired at 800 °C under  $\text{O}_2$  for 14 h before it was quenched to room temperature and moved to a glove box filled with argon.

X-ray diffraction (XRD) patterns were collected on a PANalytical X'Pert Pro equipped with Cu K $\alpha$  radiation in the  $2\theta$  range of 5–85°. All the samples were sealed with Kapton film to avoid air exposure. Profile matching of the powder diffraction data of the as-prepared  $\text{NaNi}_{1/3}\text{Co}_{1/3}\text{Fe}_{1/3}\text{O}_2$  was performed with Highscore Plus using space group R-3m.

Journal Articles  
& Patents



FastText

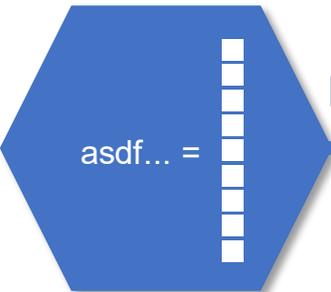
Word  
Embeddings

Annotated  
Paragraphs



Recurrent Net

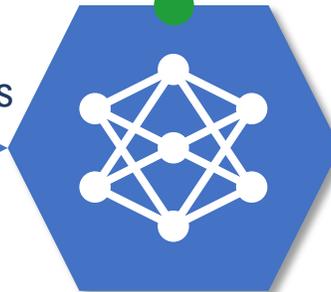
Desired  
Text



ELMo

Word  
Embeddings

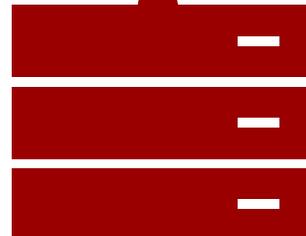
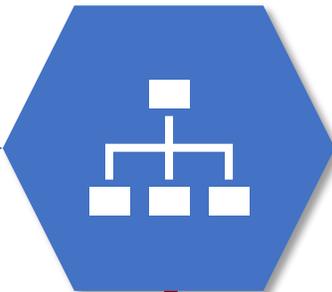
Annotated  
Sentences



Recurrent Net

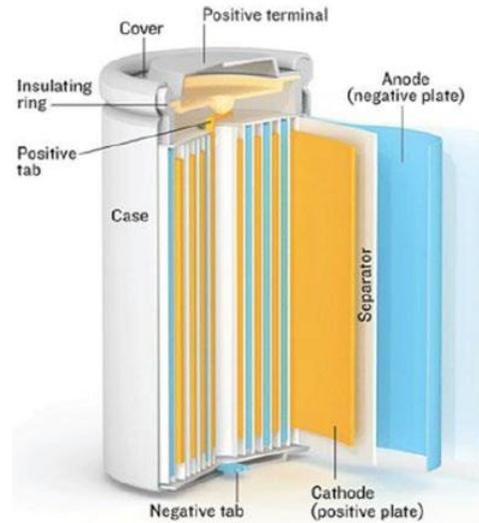
Named  
Entities

Grammatical  
Dependency  
Parser

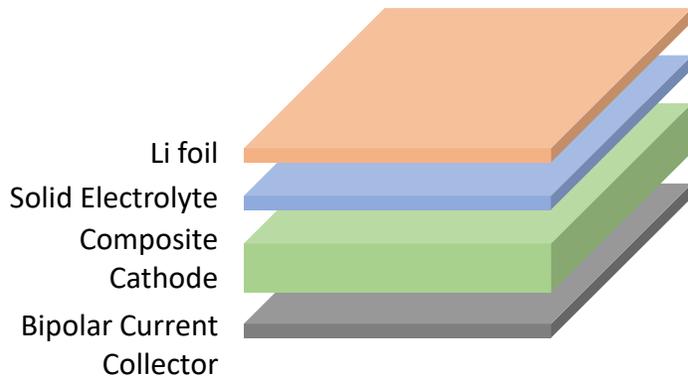
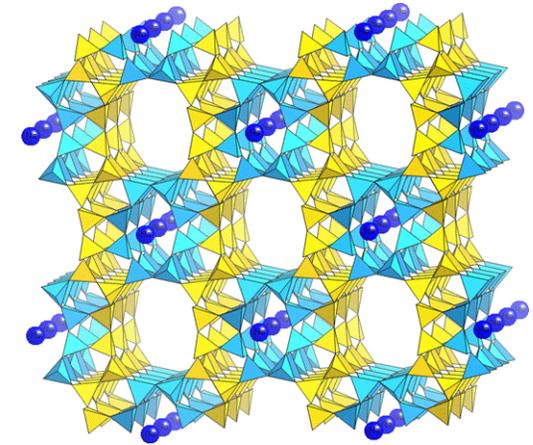


# Balance generalized approach with accuracy tailored for specific examples

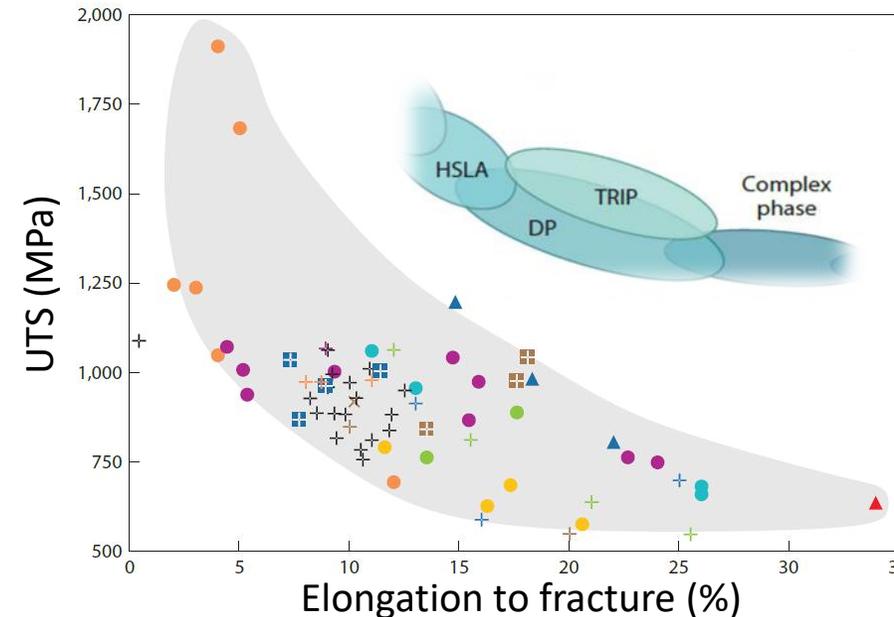
Inorganic materials synthesis



Directing zeolite crystallization

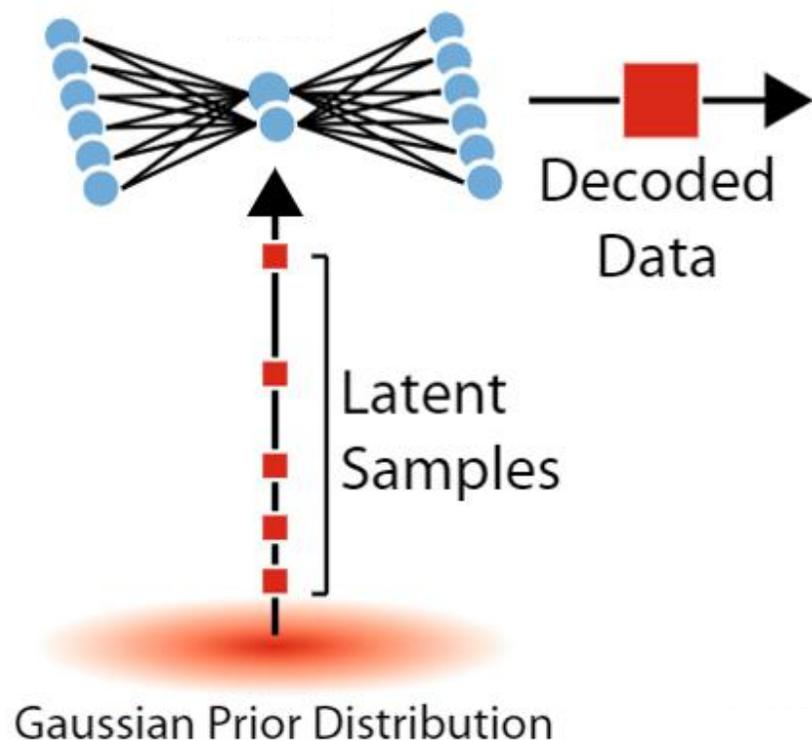


Solid state electrolyte devices



Alloy design and processing

# Time-based literature holdouts predict recipes for recently-reported perovskites



Target material	Suggested precursors	Synthesis method
InWO <sub>3</sub>	In <sub>2</sub> O <sub>3</sub> +WO <sub>2</sub>	Solid state
	InCl <sub>3</sub> +Na <sub>2</sub> WO <sub>4</sub> [1]	Solution phase
PbMoO <sub>3</sub>	PbSO <sub>4</sub> +MoCl <sub>2</sub>	Solution phase
	PbO+MoO <sub>2</sub> [2]	Solid state

**Validation includes:** reaction enthalpies, commercial availability

**Screening approach:** Within 10 attempts; at least 1 feasible route with commercially available precursors

E. Kim et al., under review

[1] J. Kamalakkannan, et al., *World Scientific News*, 2016

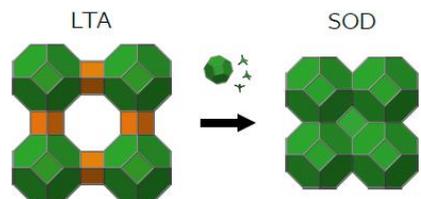
[2] H. Takatsu et al., *Physical Review B*, 2017

Collaboration with S. Jegelka



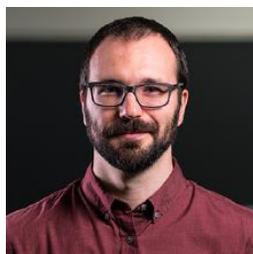
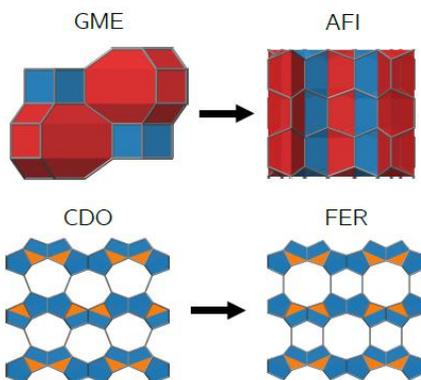
# Many more zeolites are predicted to be stable than have been realized experimentally

## Recrystallization

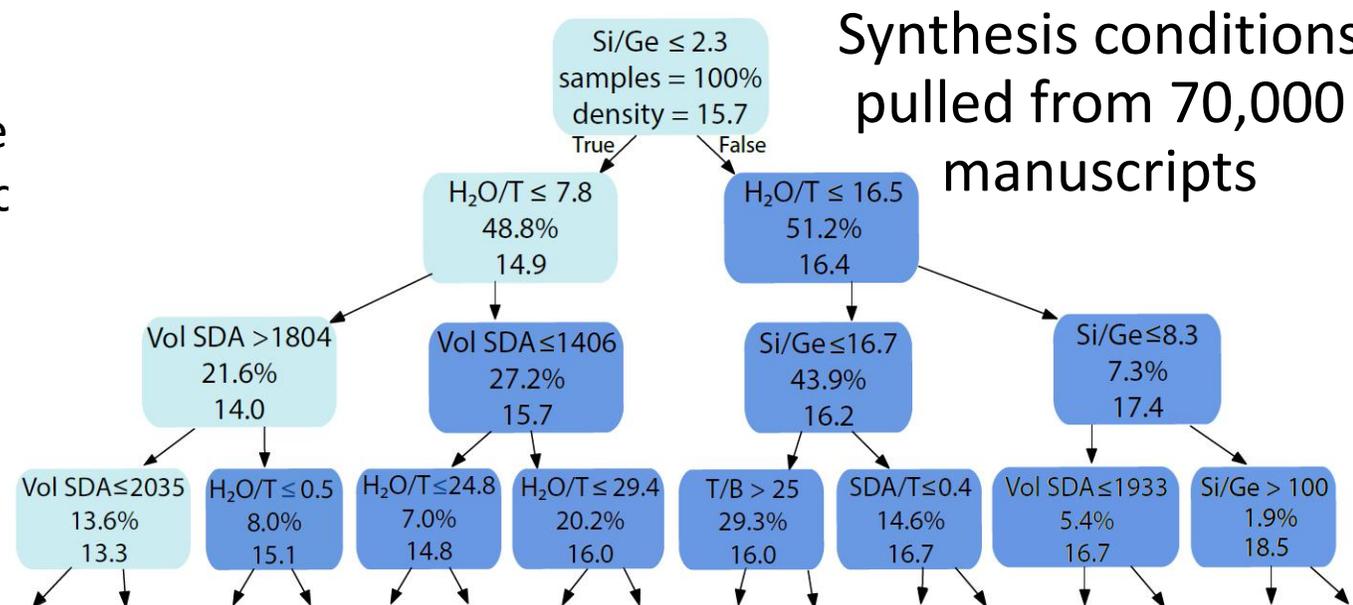


Applying similarity to 250k zeolites, we found 4k zeolite which are graph isomorphic to known ones.

## Diffusionless



with Gomez-Bombarelli group



Synthesis conditions pulled from 70,000 manuscripts

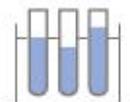
Predicting synthesis pathways for low framework density zeolites

# Production in Lab



# Manufacturing at Scale

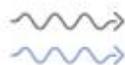
## KEY DRIVERS



Materials



Process and equipment



Heat and mass transport

## KEY OUTCOMES



Performance

## KEY DRIVERS



Materials



Process and equipment



Heat and mass transport

## KEY OUTCOMES



Performance



Reproducibility



Process rate



Process yield



Energy use



Process knowledgebase



Infrastructure maintenance

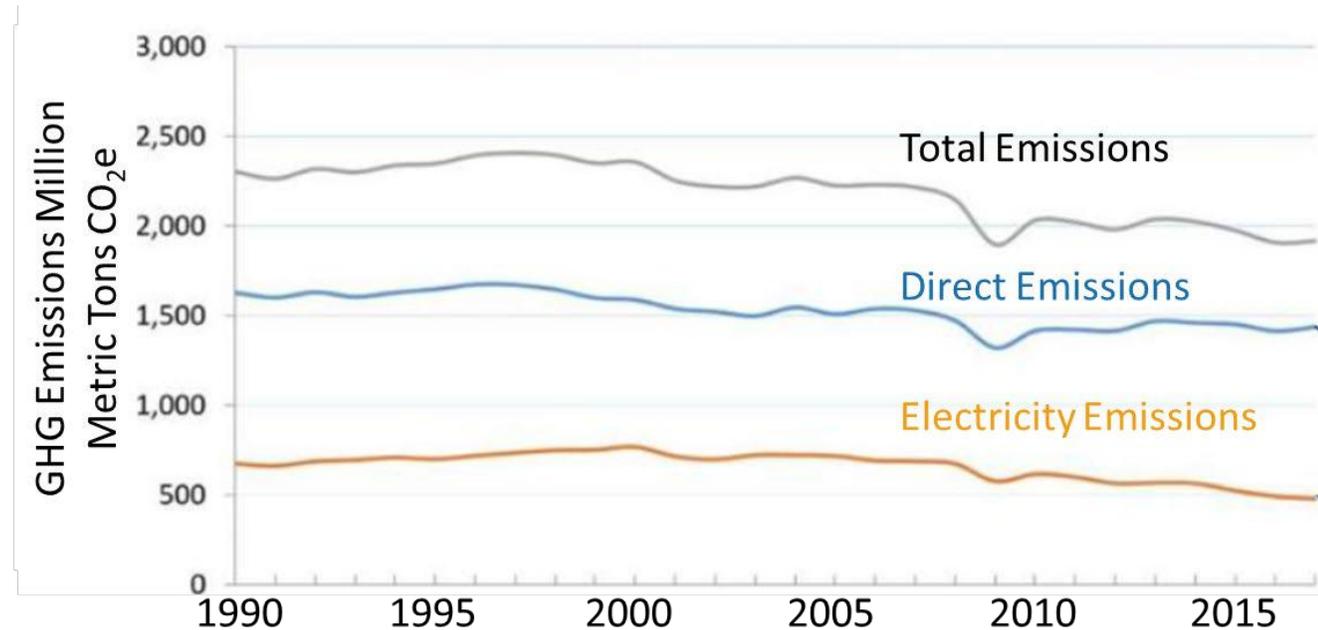
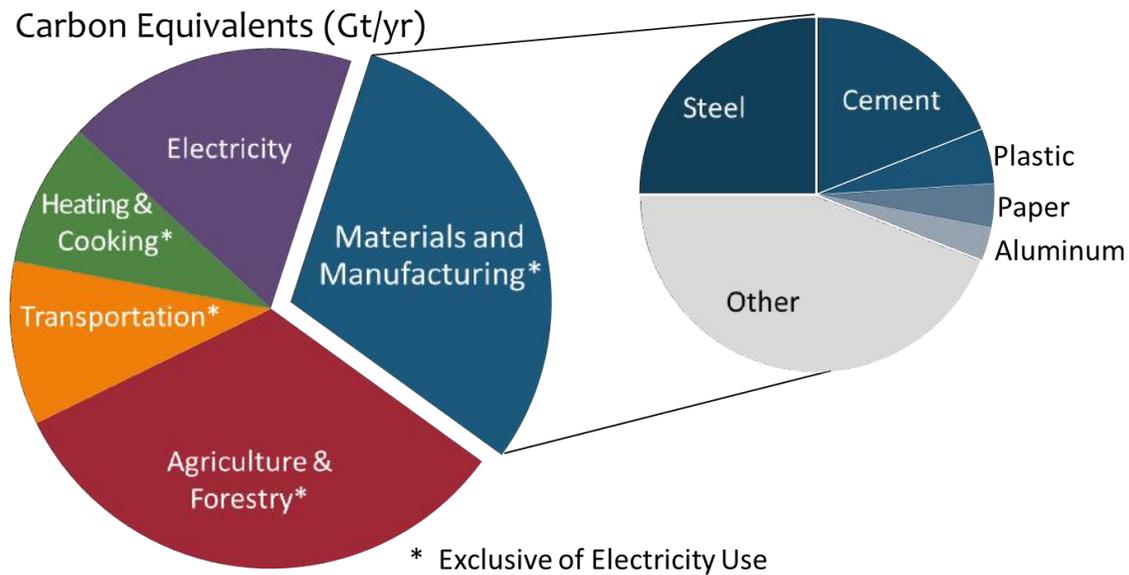


Production volume



Production cost

# There are significant opportunities beyond electricity

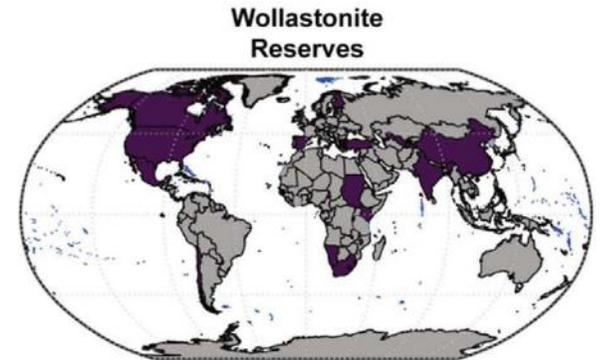
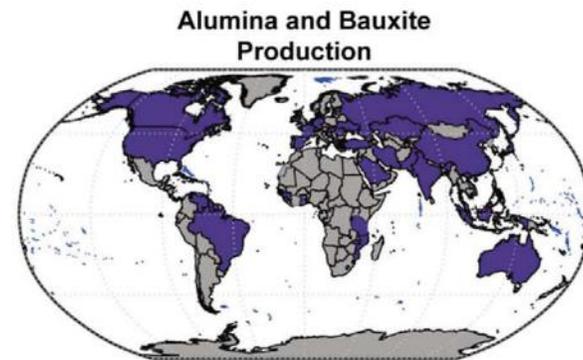
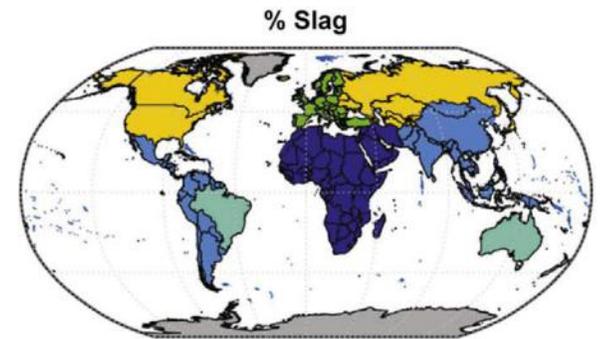
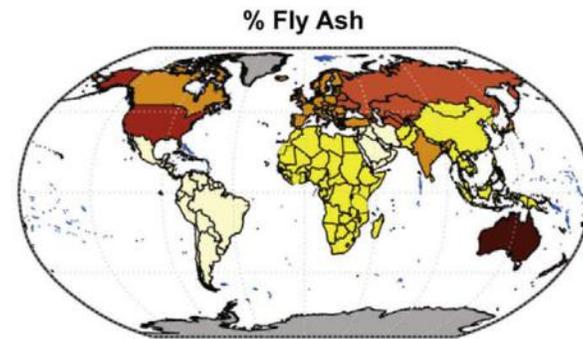
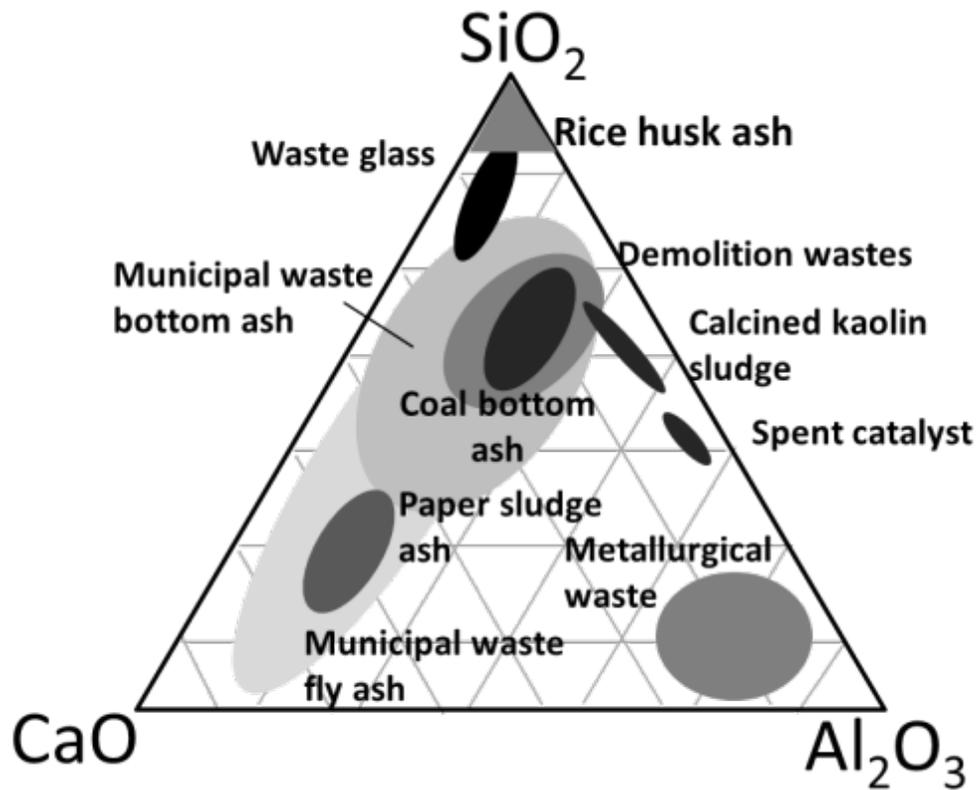


Source: DOE (2014) EPA (2018)

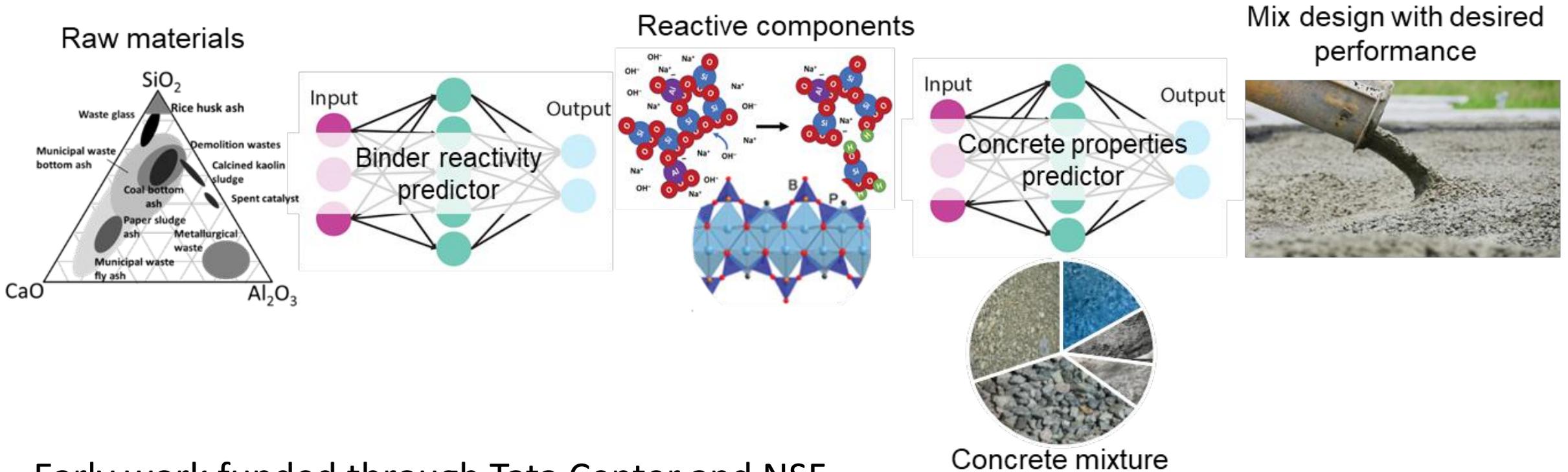
(left figure) Sassoon, R., et al.. *Quantifying the Flow of Exergy and Carbon*. 2009 MRS Spring Meeting

(right figure) Allwood, J., et al., *Options for Achieving a 50% Cut in Industrial Carbon Emissions by 2050*, ES&T, 2010, 44 (6).

# Design cements across broad set of wastes, enable sophisticated beneficial use



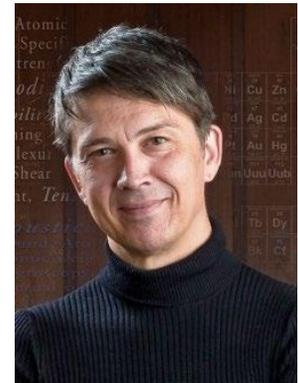
# Develop low environmental impact concrete mixtures through effective use of waste materials



Early work funded through Tata Center and NSF, now funded through MIT-IBM Watson AI Laboratory

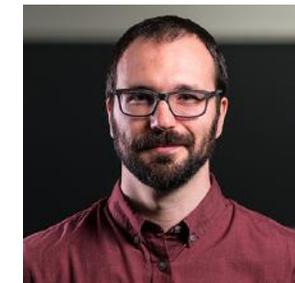
Thank you

Eddie Kim, Zach Jensen, Kevin Huang, Alex van Grootel  
Soon Kwon, Hugo Uvegi



MIT (Roman, Jegelka,  
Gomez-Bombarelli)  
Berkeley (Ceder)  
UMA (McCallum)  
ITQ (Moliner and Corma)

[olivetti.mit.edu](http://olivetti.mit.edu); [synthesisproject.org](http://synthesisproject.org)



**MATERIALS  
PROJECT**



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